Name: \_\_\_\_\_\_ Class: \_\_\_\_\_ Date: \_\_\_\_\_ ID: A

#### CP Chemistry - Zajac - Semester II Exam Study Guide / Practice Exam

#### Multiple Choice

Identify the choice that best completes the statement or answers the question.

- - a. the number of atoms of bromine in 1 mol Be,
  - b. the number of atoms of gold in 1 mol Au-
  - c. the number of molecules of nitrogen in 1 mol N2
  - d. the number of molecules of carbon mesonade in 1 mol CO
- How many moles of tangeten atoms are in 4.8 × 10<sup>70</sup> atoms of tangeten?
  - $a_c = 8.0 \times 10^2$  moles.

c.  $1.3 \times 10^{-1}$  modes

8.0 × 10<sup>3</sup> moles.

- 1.5 × 10<sup>-1</sup> modes.
- Which of the following is NOT a true about atomic mass?
  - a. The atomic mass is 12 g for magnesium.
  - The atomic mass is the mass of one mole of atoms.
  - c. The atomic mass is found by checking the periodic table.
  - The atomic mass is the number of grants of an element that is numerically equal to the mass in amu.
- 4. The mass of a mole of NaCl is the
  - a. molar mass

c. molecular mess-

b. atomic mass

d. gram atomic mass

- What is the molar mass of AuCL?
  - a. 96 a.

e. 232.5 g

b. 130 g

- d. 300.6 g
- The molar mass of C<sub>1</sub>H<sub>10</sub>, and the molar mass of CaCO<sub>1</sub> contain approximately the same number of \_
  - a. carbon atoms

at applicable

by dentiones

- d. gerantes
- 7. How many moles of Calle, are in 5.0 grams of Calle,?
  - a. 2.5 × 10<sup>-7</sup> mod

4.0 × 80° mol

b. 4.2 × 10<sup>-2</sup> good

- d. 1.0 × 10 med
- The lowest whole-number ratio of the elements in a compound is called the \_\_\_\_\_\_
  - a: empirical formula

c. bistory formula

b. molecular formula

d. representative formula

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E W Abel, F G A Stone

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topics such as credentialing legal issues legislation reimbursement and delivery systems Following each chapter are test questions which are intended to serve as an introduction to the testing arena In addition a bibliography is included for those who need a more in depth discussion of the subject matter in each chapter Organometallic Chemistry E W Abel, F G A Stone, 2007-10-31 Organometallic chemistry is an interdisciplinary science which continues to grow at a rapid pace Although there is continued interest in synthetic and structural studies the last decade has seen a growing interest in the potential of organometallic chemistry to provide answers to problems in catalysis synthetic organic chemistry and also in the development of new materials This Specialist Periodical Report aims to reflect these current interests reviewing progress in theoretical organometallic chemistry main group chemistry the lanthanides and all aspects of transition metal chemistry Specialist Periodical Reports provide systematic and detailed review coverage of progress in the major areas of chemical research Written by experts in their specialist fields the series creates a unique service for the active research chemist supplying regular critical in depth accounts of progress in particular areas of chemistry For over 80 years the Royal Society of Chemistry and its predecessor the Chemical Society have been publishing reports charting developments in chemistry which originally took the form of Annual Reports However by 1967 the whole spectrum of chemistry could no longer be contained within one volume and the series Specialist Periodical Reports was born The Annual Reports themselves still existed but were divided into two and subsequently three volumes covering Inorganic Organic and Physical Chemistry For more general coverage of the highlights in chemistry they remain a must Since that time the SPR series has altered according to the fluctuating degree of activity in various fields of chemistry Some titles have remained unchanged while others have altered their emphasis along with their titles some have been combined under a new name whereas others have had to be discontinued The current list of Specialist Periodical Reports can be seen on the inside flap of this volume **AEC** Reports Declassified ,1957 Nuclear Magnetic Resonance G A Webb, 2007-10-31 As a spectroscopic method Nuclear Magnetic Resonance NMR has seen spectacular growth over the past two decades both as a technique and in its applications Today the applications of NMR span a wide range of scientific disciplines from physics to biology to medicine Each volume of Nuclear Magnetic Resonance comprises a combination of annual and biennial reports which together provide comprehensive of the literature on this topic This Specialist Periodical Report reflects the growing volume of published work involving NMR techniques and applications in particular NMR of natural macromolecules which is covered in two reports NMR of Proteins and Acids and NMR of Carbohydrates Lipids and Membranes For those wanting to become rapidly acquainted with specific areas of NMR this title provides unrivalled scope of coverage Seasoned practitioners of NMR will find this an in valuable source of current methods and applications Specialist Periodical Reports provide systematic and detailed review coverage in major areas of chemical research Compiled by teams of leading authorities in the relevant subject areas the series creates a unique service for the active research chemist with regular in depth accounts of progress in particular fields of chemistry

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Fundamentals and Applications provide both the theoretical foundation as well as the practical aspects <u>Chemistry Education</u> Javier García-Martínez, Elena Serrano-Torregrosa, 2015-02-17 Winner of the CHOICE Outstanding Academic Title 2017 Award This comprehensive collection of top level contributions provides a thorough review of the vibrant field of chemistry education Highly experienced chemistry professors and education experts cover the latest developments in chemistry learning and teaching as well as the pivotal role of chemistry for shaping a more sustainable future Adopting a practice oriented approach the current challenges and opportunities posed by chemistry education are critically discussed highlighting the pitfalls that can occur in teaching chemistry and how to circumvent them The main topics discussed include best practices project based education blended learning and the role of technology including e learning and science visualization Hands on recommendations on how to optimally implement innovative strategies of teaching chemistry at university and high school levels make this book an essential resource for anybody interested in either teaching or learning chemistry more effectively from experience chemistry professors to secondary school teachers from educators with no formal training in didactics to frustrated chemistry students

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