



CHEMISTRY

CHEAT SHEET

EQUILIBRIUM

EQUILIBRIUM CHEAT SHEET

Chemical Equilibrium occurs when the concentrations of the chemical species remain the same with time (dynamic equilibrium). Equilibrium reactions are represented by the symbol \rightleftharpoons .

General Equilibrium		
<p>General Equilibrium:</p> <p>Chemical Equilibrium occurs when the concentrations of the chemical species remain the same with time (dynamic equilibrium). Equilibrium reactions are represented by the symbol \rightleftharpoons.</p> <p>Equilibrium Constant (K_c)</p> <p>For a general reaction:</p> $aA + bB \rightleftharpoons cC + dD$ <p>The equilibrium constant is given by:</p> $K_c = \frac{[C]^c [D]^d}{[A]^a [B]^b}$	<p>Equilibrium Constant (K_c)</p> <p>For a general reaction:</p> $aA + bB \rightleftharpoons cC + dD$ <p>The equilibrium constant is given by:</p> $K_c = \frac{[C]^c [D]^d}{[A]^a [B]^b}$	<p>Graphical Representation of Equilibrium</p>
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Le Chatelier's Principle: the effect of changing conditions on equilibrium		
<p>Le Chatelier's Principle:</p> <p>If a system at equilibrium is subjected to a change in conditions, the system will adjust itself to counteract the change and restore equilibrium.</p>	<p>Le Chatelier's Principle:</p> <p>If a system at equilibrium is subjected to a change in conditions, the system will adjust itself to counteract the change and restore equilibrium.</p>	<p>Le Chatelier's Principle:</p> <p>If a system at equilibrium is subjected to a change in conditions, the system will adjust itself to counteract the change and restore equilibrium.</p>

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