Acids, Bases & pH Review - Honors Chemistry

Directions: For all multiple choice questions, highlight the correct answer(s). Answer short answer questions in the box provided.

Part 1: Properties & definition of Acids & Bases

Complete the table below.

	ACIOS	BASES
Taxte	7	5
pH range	- 2	
Conducting electricity	70	
Neutralize		
Litmus	37	
Phenoiphthalein		

- 2. An indicator
 - a. Its a type of acid.
 - b. is a type of base.
 - c. has a neutral pH.
 - d. changes color depending on the pH.
- 3. Littinus and phenolofithalein are indicators. What does that meanly
 - Acids and bases "dissociate" in water to form H" (for acids) and OH (for bases).
 What does "dissociate" mean in this context?
 - Identify whether each substance is an acid or a base.
 - a. Hitter i.
 - B. HICHIONS
 - c. Alloomily:
 - d. NH + HyO + NH + OH; NH in add):
 - H,SO, + H,O → H,O* + HSO, : H,SO, is sónit
 - f. HNO, H* + NO, : HNO, is admi-

Chemistry Acids And Bases Answers Study Guide

Ying Liu

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