

BALANCING CHEMICAL EQUATIONS

Load the simulation *Balancing Chemical Equations*. This link is in your JupiterGrades messages <http://phet.colorado.edu/en/simulation/balancing-chemical-equations>

CRITICAL THINKING QUESTIONS

REVIEW:

MOLECULES: A GROUP OF ATOMS BONDED TOGETHER

ATOMS: THE BASIC UNIT OF A CHEMICAL ELEMENT

- Explore the *Balancing Chemical Equations* simulation. Discuss with your group what you find.
 - What are the tools you can use in the simulation to tell you when an equation is balanced?
Scale or bar graph

- For each balanced reaction, indicate the total number of molecules in the table below.

Reaction	Total Number of Molecules	
	Reactant Side (Left)	Product Side (Right)
Make Ammonia	4	2
Separate Water	2	3
Combust Methane	3	3

- Is the number of total molecules on the left side of a balanced equation always equal to the number of total molecules on the right side of the equation? Explain your answer.
No because the atoms rearrange in a chemical reaction to form new molecules.

- For each balanced reaction, indicate the total number of atoms in the table below.

Reaction	Total Number of Atoms	
	Reactant Side (Left)	Product Side (Right)
Make Ammonia	8	8
Separate Water	6	6
Combust Methane	9	9

- Is the number of total atoms on the left side of a balanced equation always equal to the number of total atoms on the right side of the equation?
Yes
- What is the same on the left and right side of a balanced equation? Explain your answer.
The numbers are all the same because they're balanced.
- As a group, play level 1 of the balancing equation game. Write down the strategies your group uses to balance chemical equations.

Chemistry A Guided Inquiry Balancing Chemical Equations

Festo Kayima



Chemistry A Guided Inquiry Balancing Chemical Equations:

Introductory Chemistry Michael P. Garoutte, Ashley B. Mahoney, 2015-08-10 The ChemActivities found in Introductory Chemistry A Guided Inquiry use the classroom guided inquiry approach and provide an excellent accompaniment to any one semester Introductory text Designed to support Process Oriented Guided Inquiry Learning POGIL these materials provide a variety of ways to promote a student focused active classroom that range from cooperative learning to active student participation in a more traditional setting

Preparing for Chemistry Teaching Festo Kayima, 2025-08-13 This textbook is a comprehensive chemistry didactics resource for chemistry teacher educators chemistry teachers and trainees It provides research grounded and practical based pedagogical experiences examples and frameworks for chemistry teachers as well as a foundation for planning and implementing productive chemistry lessons The book provides a conceptual and practical roadmap illuminating which didactic knowledge elements are relevant for becoming a chemistry teacher The book starts off with a pedagogically laden however experience based justification for the relevance of chemistry didactics and then progressively breaks down the different knowledge elements that form a complete set of the didactic knowledge and skill elements a teacher needs for teaching Concrete examples are provided to allow the reader to operationalize the ideas and concepts presented in the book The structure of the chapters enables the reader to engage progressively and actively with its contents and provided examples allowing a deep understanding of the diverse links between the presented topics forming a complete set of the didactic knowledge and skills relevant for successful chemistry teaching

General, Organic, and Biological Chemistry Michael P. Garoutte, 2014-02-24 Classroom activities to support a General Organic and Biological Chemistry text Students can follow a guided inquiry approach as they learn chemistry in the classroom General Organic and Biological Chemistry A Guided Inquiry serves as an accompaniment to a GOB Chemistry text It can suit the one or two semester course This supplemental text supports Process Oriented Guided Inquiry Learning POGIL which is a student focused group learning philosophy of instruction The materials offer ways to promote a student centered science classroom with activities The goal is for students to gain a greater understanding of chemistry through exploration

Chemistry Richard S. Moog, John J. Farrell, 2017-06-26 In the newly updated 7th Edition Chemistry A Guided Inquiry continues to follow the underlying principles developed by years of extensive research on how students learn and draws on testing by those using the POGIL methodology This text follows the principles of inquiry based learning and correspondingly emphasizes underlying chemistry concepts and the reasoning behind them This text provides an approach that follows modern cognitive learning principles by having students learn how to create knowledge based on experimental data and how to test that knowledge

Learning and Performance Assessment: Concepts, Methodologies, Tools, and Applications Management Association, Information Resources, 2019-10-11 As teaching strategies continue to change and evolve and technology use in classrooms continues to increase it is imperative that their impact on student learning is monitored and assessed New

practices are being developed to enhance students participation especially in their own assessment be it through peer review reflective assessment the introduction of new technologies or other novel solutions Educators must remain up to date on the latest methods of evaluation and performance measurement techniques to ensure that their students excel Learning and Performance Assessment Concepts Methodologies Tools and Applications is a vital reference source that examines emerging perspectives on the theoretical and practical aspects of learning and performance based assessment techniques and applications within educational settings Highlighting a range of topics such as learning outcomes assessment design and peer assessment this multi volume book is ideally designed for educators administrative officials principals deans instructional designers school boards academicians researchers and education students seeking coverage on an educator s role in evaluation design and analyses of evaluation methods and outcomes *Handbook of Research on Innovative Pedagogies and Technologies for Online Learning in Higher Education* Vu, Phu,Fredrickson, Scott,Moore, Carl,2016-12-28 The integration of technology has become an integral part of the educational environment By developing new methods of online learning students can be further aided in reaching goals and effectively solving problems The Handbook of Research on Innovative Pedagogies and Technologies for Online Learning in Higher Education is an authoritative reference source for the latest scholarly research on the implementation of instructional strategies tools and innovations in online learning environments Featuring extensive coverage across a range of relevant perspectives and topics such as social constructivism collaborative learning and projects and virtual worlds this publication is ideally designed for academicians practitioners and researchers seeking current research on best methods to effectively incorporate technology into the learning environment

Process Oriented Guided Inquiry Learning (POGIL) Richard Samuel Moog,2008 POGIL is a student centered group learning pedagogy based on current learning theory This volume describes POGIL s theoretical basis its implementations in diverse environments and evaluation of student outcomes *Science Units for Grades 9-12* Randy L. Bell,Joe Garofalo,2005 Sample topics include cell division virtual dissection earthquake modeling the Doppler Effect and more **The Big Book of Chemistry Teacher Stories** Jeff Lark, Stories from years of teaching high school chemistry *Science Spectrum* Holt Rinehart & Winston,Holt, Rinehart and Winston Staff,2003-03 **Resources in Education** ,1985 *ChemDiscovery Teacher Edition* Olga I. Agapova,2002 *Science Education* Keith S. Taber,Ben Akpan,2016-12-27 This book comprises a wide range of scholarly essays introducing readers to key topics and issues in science education Science education has become a well established field in its own right with a vast literature and many active areas of scholarship Science Education An International Course Companion offers an entry point for students seeking a sound but introductory understanding of the key perspectives and areas of thinking in science education Each account is self contained and offers a scholarly and research informed introduction to a particular topic theme or perspective with both citations to key literature and recommendations for more advanced reading Science Education An International Course Companion allows readers such as

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Donaldson, Kathleen S. Lundgren, Howard Marget Spiro, 2003-01-01 Presents a collection of first person accounts of what life is like in the medical field *Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics*

Earth Science, 2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction **Geology at Every Scale** Annette

Summers Engel, Robert D. Hatcher Jr., 2018-03-30 The chapters in this guidebook are organized according to major geologic themes starting first with field trips in the Knoxville area that highlight in some way local carbonates and then by ending with field trips focused on regional tectonics that include travel to North and South Carolina and Georgia **ChemDiscovery**

Student Guide Olga I. Agapova, 2002 **Balancing Chemical Equations Worksheet** Crispin Collins, 2020-09-12

Struggling with balancing chemical reaction Balancing chemical equations can look intimidating for lot of us The good news is that practice makes perfect Master balancing skill with this workbook packed with hundreds of practice problems This book is for anyone who wants to master the art of balancing chemical reactions First few chapters of this book are step by step explanation of the concepts and other chapters are for practicing problems This book help students develop fluency in balancing chemical equation which provides plenty of practice Methods to solve with the explanation Total of 550 problems to solve with answer key 450 chemical reactions to practice with answer key 100 practice problems that are needed before balancing a chemical reaction with answer key Click the Buy now button to take advantage of this book to help yourself in mastering balancing skill **The World of Science Education** Femi S. Otulaja, Meshach B. Ogunniyi, 2017-09-12 Each

volume in the 7 volume series The World of Science Education reviews research in a key region of the world These regions include North America South and Latin America Asia Australia and New Zealand Europe and Israel North Africa and the Middle East and Sub Saharan Africa The focus of this Handbook is on research in science education in mostly former British colonies in Sub Saharan Africa and the scholarship that most closely support this program The reviews of the research situate what has been accomplished within a given field in Sub Saharan Africa rather than an international context The purpose therefore is to articulate and exhibit regional networks and trends that produced specific forms of science education

The thrust lies in identifying the roots of research programs and sketching trajectories focusing the changing face of problems and solutions within regional contexts. The approach allows readers to review what has been done and accomplished, what is missing, and what might be done next.

Lavoisier—the Crucial Year Henry Guerlac, 2019-05-15

The author explores the origins of the eighteenth century chemical revolution as it centers on Antoine Laurent Lavoisier's earliest work on combustion. He shows that the main lines of Lavoisier's theory, including his theory of a heat fluid caloric, were elaborated well before his discovery of the role played by oxygen. Contrary to the opinion prevailing at that time, Lavoisier suspected and demonstrated by experiment that common air, or some portion of it, combines with substances when they are burned. Professor Guerlac examines critically the theories of other historians of science concerning these first experiments and tries to unravel the influences which French, German, and British chemists may have had on Lavoisier. He has made use of newly discovered material on this phase of Lavoisier's career and includes an appendix in which the essential documents are printed together for the first time.

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