

## Chemistry 30

## Unit 3: Chemical Equilibrium

## Assignment 3: 3-1 to 3-5

Le Chatelier's Principle

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1. State Le Chatelier's Principle.

2. What are three stresses that can affect the position of an equilibrium?

Identify the one stress that will cause the value of  $K_{eq}$  to change.

3. State the effect of a catalyst on equilibrium.

4. Methanol (methyl alcohol,  $\text{CH}_3\text{OH}$ ) can be manufactured using the following equilibrium reaction:



Predict the effect of the following changes on the equilibrium concentration of  $\text{CH}_3\text{OH(g)}$ . Will its concentration increase, decrease, or remain the same?

a) The temperature of the system is decreased.

b) The pressure of the system is increased.

c) More  $\text{H}_2\text{(g)}$  is added.

d) A catalyst is added to the system.

# Chemistry 30 Unit 3 Chemical Equilibrium

**Annelies Wilder-Smith**



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