

Q3. Use the periodic table to categorize the following elements as metals (M) or nonmetals (NM).

S ..... P ..... Cr ..... Ni ..... Sr ..... I .....

Q4. A neutral atom has a mass number of 56 and 30 neutrons. Write its isotopic symbol.

Q5. Determine the number of protons, electrons, and neutrons of the following atoms.

Al-27

U-238

Cu-64

Bc-80

Ar-40

Q6. Describe the location (group and period) of the following metals and nonmetals on the periodic table.

Zn

I

Fr

Kr

Q7. Write the isotope symbol that corresponds to the following subatomic particles.

A)  $P = 17$ ,  $E = 17$ , AND  $N = 20$

B)  $P = 36$ ,  $E = 36$ , AND  $N = 48$

C)  $P = 26$ ,  $E = 26$ , AND  $N = 30$

D)  $P = 1$ ,  $E = 1$ , AND  $N = 0$

### FINAL CHECKLIST

1. Clean your equipment and work benches before you leave.
2. Return all equipment to their proper storage area.
3. Submit your answers to the review questions along with your technical report to your instructor before the next laboratory session.

# Chemistry 101 Lab Manual Questions

**Sasha Hu**



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