Study Guide Unit 9

- KNOW the following vocabulary words (this means you NEED to be able to DEFINE them ALL):
- a. Solution- m. Solute-
- b. Homogenous-

n. Solvent-

c. Heterogeneous-

o. Acid-

d. Soluble-

p. Base-

e. Insoluble-

q. Titration-

f. Miscible-

r. Analyte-

g. Immiscible-

s. Titrant-

h. Solubility-

t. Monoprotic acid-

i. Allloy-

u. Diprotic Acid

j. Saturated-

v. Triprotic Acid

- k. Unsaturated-
- I. Supersaturated-
- 2. How is the solubility of a gas affected by temp. and pressure?

Pressure up = solubility up

Temp up = solubility down

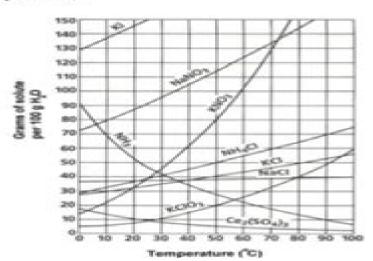
- Under what conditions are salts most soluble?High temperatures
- Use the solubility curve to the right to answer the following.
- a. What mass of KNO₃ will make a saturated solution at 60°C? 100g
- b. Which of the compounds shown on the solubility curve decrease solubility as temperature increases? NH3, Ce2(SO4)3
- c. What type of solution do I have if I add 110 g of NaNO₃ in 100g of 20°C water?

Saturated

- d. What mass of KCl can go into solution at 80°C? 50 e
- e. What is the least soluble compound at 20°C?

KCIO3

f. Excluding KI, what is the most soluble compound at 80°C? KNO3



Chemistry Study Guide Answers What Are Solutions

Yijin Wang

Chemistry Study Guide Answers What Are Solutions:

Chemistry Study Guide Answers What Are Solutions Book Review: Unveiling the Magic of Language

In a digital era where connections and knowledge reign supreme, the enchanting power of language has be much more apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Chemistry Study Guide Answers What Are Solutions**," compiled by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound impact on our existence. Throughout this critique, we will delve in to the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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