

Chemistry 530 Midterm Review

Exam on Monday Dec 16 at 8:30

Section I = 75 pts

- Consists of 75 multiple choice questions worth 1 pt. each
- Questions cover all material covered during the second semester
- Questions will be similar to multiple choice problems in the book and review guides on-line
- Practicing the review guides is the best approach; answers will be posted on-line

Section II = 95 pts

- Consists of 19 short answer problems totaling 95 points
- Questions cover the whole semester
- Examples of types of questions
 - Metric Conversions
 - Volume Calculations
 - Percent Composition
 - Atomic Mass
 - Electron configuration
 - Nomenclature
 - Balancing Equations
 - Predicting the products of a reactions
 - Identifying types of reactions
 - Mole Conversions
 - Molar Mass of a compound
 - Molar Mass of a gas at STP
 - Density of a gas at STP
 - Empirical & Molecular Formula

Study approach

- First, read through your notes. Once thoroughly and once more quickly.
- Go through the review packet and make sure you are comfortable with all of the terms and how to do problems
- Do the review packets for help with multiple choice
- Do questions on old tests for practicing written problems
- Make sure you have practiced AT LEAST one of the types of example problems

Chapter 1 – Introduction to Chemistry

- Chemistry: the branch of science that studies the composition and properties of matter.
- Scientific Method:
 - Observation
 - Hypothesis
 - Controlled Experiment
- Theory versus Natural Law

Prerequisite Science Skills

- Rules for Significant Figures:
 - 1) All non-zero digits are significant.
 - 2) Zeros are significant only when they are *between* two non-zero digits or if they occur *after* a decimal point and *after* a non-zero digit.
 - 3) In *multiplication or division*, you cannot end up with more significant figures than you started with.
 - 4) In *addition or subtraction*, you cannot have more digits to the *right* of the decimal than you started with.
- Exponential numbers
- Scientific Notation = $D.DD \times 10^n$ where $1.00 \leq D.DD \leq 9.99$

Chemistry Second Semester Study Guide

Sebastian Brünink



Chemistry Second Semester Study Guide:

Differentiated Study Guide High School Chemistry Carl David Morneweck, Chester Arthur Buckner, Gale R. Kirschner, 1935 *Catalog of Copyright Entries. Third Series* Library of Congress. Copyright Office, 1976 **College Chemistry II** Norman H. Nachtrieb, 1960 **State Manual of the Courses of Study for the High Schools of Oregon ...** Oregon. Office of Superintendent of Public Instruction, 1919 **Journal of Chemical Education**, 1925 Includes Report of New England Association of Chemistry Teachers and Proceedings of the Pacific Southwest Association of Chemistry Teachers

State Curriculum Guides for Science, Mathematics, and Modern Foreign Languages, a Bibliography, 1960 *Chemical Energetics, Equilibria and Functional Group Organic Chemistry I - Laboratory* Mr. Rohit Manglik, 2024-03-03

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based development is needed to fully capitalize on this potential Learning Science will guide academic researchers developers publishers and entrepreneurs from the digital simulation and gaming community and education practitioners and policy makers toward the formation of research and development partnerships that will facilitate rich intellectual collaboration Industry government agencies and foundations will play a significant role through start up and ongoing support to ensure that digital games and simulations will not only excite and entertain but also motivate and educate Catalog University of Wisconsin, 1895 Some nos include Announcement of courses **Monthly Catalog of United States Government Publications**, 1994 **Directory of Distance Learning Opportunities** Modoc Press, Inc., 2003-02-28 This book provides an overview of current K 12 courses and programs offered in the United States as correspondence study or via such electronic delivery systems as satellite cable or the Internet The Directory includes over 6 000 courses offered by 154 institutions or distance learning consortium members Following an introduction that describes existing practices and delivery methods the Directory offers three indexes Subject Index of Courses Offered by Level Course Level Index Geographic Index All information was supplied by the institutions Entries include current contact information a description of the institution and the courses offered grade level and admission information tuition and fee information enrollment periods delivery information equipment requirements credit and grading information library services and accreditation

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