

Chemistry of the Elements

SECOND EDITION

N N Greenwood • A Earnshaw

School of Chemistry, University of Leeds, U. K.

When this innovative textbook first appeared in 1984 it rapidly became a great success throughout the world and has already been translated into several. European and Asian languages. Now the authors have completely revised and updated the text, including more than 2000 new literature references to work published since the first edition. No page has been left unaffered but the novel features which proved so attractive have been retained. The book presents a balanced, coherent and comprehensive account of the chemistry of the elements for both undergraduate and postgraduate students. This crucial central area of chemistry is full of ingenious experiments, intriguing compounds and exciting new discoveries. The authors specifically avoid the term inorganic chemistry since this evokes an outmoded view of chemistry which is no longer appropriate in the final decade of the 20th century.

Accordingly, the book covers not only the 'inorganic' chemistry of the elements, but also analytical, theoretical, industrial, arganometallic, bio inorganic and other cognate areas of chemistry. The authors have broken with recent tradition in the teaching of their subject and adopted a new and highly successful approach based on descriptive chemistry. The chemistry of the elements is still discussed within the content of an underlying theoretical framework, giring cohesion and structure to the text, but at all times the chemical facts are emphasized. Students are invited to enter the exciting world of chemical phenomena with a sound knowledge and understanding of the subject, to approach experimentation with an open mind, and to assess observations reliably. This is a book that students will not only value during. Their formal education, but will keep and reter to throughout their convers as chemists.

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CONTENTS: Origin of the elements, isotopes and atomic weights. Chemical periodicity and the periodic table. Hydrogen, Lithium, sodium, polassium, rubidium, adesium and francium, Beryllium, magnesium, calcium, strontium, barium and radium, Boron, Aluminium, gallium, indium and thallium, Carbon, Silicon, Germanium, tin and lead. Nitrogen: Phosphorus, Arsenic, antimony and blamuth; Grygen: Sulfur, Selenium, tellurium and polanium. The hologens, fluorine, chlorine, bromine, ladine and astatine: The noble gases; helium, neon, argon, krypton, xemon, and radion; Coordination and organometallic compounds; Scandium, yllrium, lanthonum and actinium. Thanium, pirconium and hafnium. Yanadium, nicolium and tantalum. Chromium, malybdenium and tungiten. Hanganese, technetium and menium; Iron, ruthenium and assistine. Cabat, rhadium and iridium. Hickel, polladium, and platinum; Copper, sitrer and gold. Zinc, cadmium and mencury. The lanthonide elements:

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distinctions taken into account when classifying the elements in the periodic table The alkali metals for example are metals but have such special properties that they are given their own classification The same is true for the alkaline earths Alkali and Alkaline Earth Metals Second Edition presents the current scientific understanding of the physics chemistry geology and biology of these two families of elements including how they are synthesized in the universe when and how they were discovered and where they are found on Earth With information pertaining to the discovery and naming of these elements as well as new developments and dilemmas this newly updated eBook examines how humans use alkalis and alkaline earths and their benefits and challenges to society health and the environment Lithium sodium potassium magnesium and calcium are only a few of the topics covered in this full color resource Alkali and Alkaline Earth Metals Second Edition provides students and scientists with an up to date understanding of each of the nonmetals where they came from how they fit into our current technological society and where they may lead us Transition Metals, Second Edition Monica Halka, Brian Nordstrom, 2019-12-01 More so than any of the other major groups of elements in the periodic table the transition metals have shaped human history and have been the workhorses of industry The discovery of metallic copper ended the Stone Age and ushered in the Bronze Age Alloys of iron especially steel later took over and the Iron Age replaced the Bronze Age Copper silver and gold and more recently platinum have been the precious metals from which coins and jewelry have been made from ancient times to the present Each chapter in the newly updated full color Transition Metals Second Edition discusses a group of elements including their similarities and differences and current research and applications Ideal for high school or college students interested in chemistry and physics this straightforward resource is devoted to the chemical and physical properties of transition metals and how they are useful in everyday life Some of the transition metals covered include scandium titanium manganese cobalt and zinc Nonmetals, Second Edition Monica Halka, Brian Nordstrom, 2019-12-01 Praise for the previous edition a solid addition to a high school and public library science collection Recommended Library Media Connection Materials that are poor conductors of electricity are generally considered nonmetals One important use of nonmetals is the ability to insulate against current flow The Earth's atmosphere is composed of nonmetallic elements but lightning can break down the electron bonds and allow huge voltages to make their way to the ground Water in its pure form is nonmetallic though it almost always contains impurities called electrolytes that allow for an electric field With an exploration of the benefits and challenges to society health and the environment Nonmetals Second Edition provides readers with new developments in the research of nonmetals including where they came from how they fit into our current technological society and where they may lead us Written in an easy to read format this newly updated full color resource discusses new developments and dilemmas past present and future uses of nonmetals in science and technology and much more Nonmetals explored in this title include hydrogen carbon nitrogen phosphorus oxygen sulfur and selenium Halogens and Noble Gases, Second Edition Monica Halka, Brian Nordstrom, 2019-12-01 In spite of their

adjacency in the periodic table halogens and nonmetals have very different properties Halogens are among the most chemically reactive elements in the periodic table exhibiting a diverse chemistry in terms of the large numbers of compounds they can form On the other hand noble gases are the least chemically reactive elements In fact before the 1960s chemists referred to these elements as inert gases because it was believed that they exhibited no chemistry whatsoever Providing the basics of these elements including their role in history and some of the important scientists involved in their discovery this newly updated full color resource features up to date scientific understanding in a clear and accessible format Halogens and Noble Gases Second Edition examines the ways humans use halogens and noble gases and the resulting benefits and challenges to society health and the environment Fluorine chlorine bromine iodine helium and krypton are covered in this eBook along with the fundamentals of chemistry and physics as well as possible future developments in halogen and noble gas science and its applications Cyclopedia of American Agriculture Liberty Hyde Bailey, 1909 Practical Handbook of Soil, Vadose Zone, and Ground-Water Contamination J. Russell Boulding, Jon S. Ginn, 2016-04-19 A synthesis of years of interdisciplinary research and practice the second edition of this bestseller continues to serve as a primary resource for information on the assessment remediation and control of contamination on and below the ground surface Practical Handbook of Soil Vadose Zone and Ground Water Contamination Assessment Prev **Iournal** Missouri State Medical Association, 1911 The Athenaeum ,1890 Quarterly literary advertiser ,1890 Catalogue of the Science Library in the South Kensington Museum South Kensington Museum. Science Library, 1891 Van Nostrand's Chemical Annual .1909 The issues for 1907 and 1909 contain a Review of chemical literature The London Medical Record .1885 Catalogue of the Educational Division of the South Kensington Museum ,1876 Catalogue of the educational division of **Educational Times** ,1884 the South Kensington museum Victoria and Albert museum, 1876 Catalogue of the Library of the Pharmaceutical Society of Great Britain Pharmaceutical Society of Great Britain. Library, 1853 The

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