# 6. The Chemistry Times published a Lewis Logic Puzzle!

Use the following clues to assign numbers to the 10 elements listed in the top row, You may either use the table to mark a check for the correct matchings or write out the matchings separately.

- a. Element 3 is one period above element 8 in the periodic table, and they are both in the same group of the periodic table.
- b. Element 2 is not boron.
- c. Element 4 can have a 1/4 full octet, while element 5 also commonly has an incomplete octet.
- d. Element 2 can participate in ionic or covalent interactions, and is a metal not in the alkali metals.

B

- e. Element 3 is the most electronegative element of the 10 listed.
- Element 1 does not form covalent molecules. f.
- In the conjugate base of phosphoric acid (this means H.PO.), element 6 carries the negative charge in all resonance structures, while element 7 obtains partial positive E. character (is partially positive in charge)
- Element 8 has the largest electron affinity of the 10 elements listed
- Element 9 usually bonds to more other atoms in a molecule than element 10.

i.	Elemen	t 9 usuai	ly bonds	to more		E	Al	CI	Na	P
	Н	В	C	N	0	1	1100	No. of the last	111111	11 2721111
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**Ying Liu** 

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