

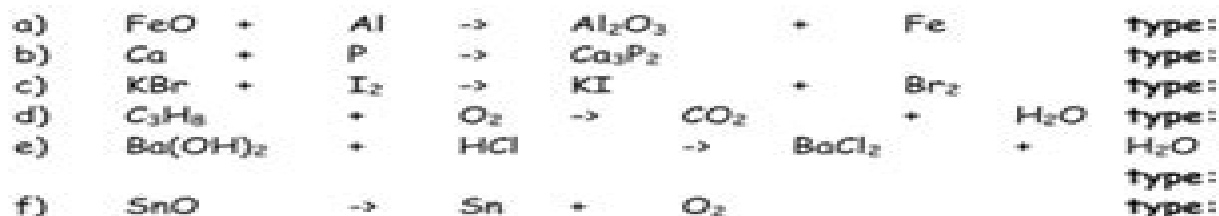
Grade 10 Chemistry Review

1. Use the periodic table to answer the following questions concerning calcium.
 - a) In what period is it located?
 - b) In what group is it located?
 - c) What is its atomic number?
 - d) What is its atomic mass?
 - e) In what state would it be found at room temperature?
2. Explain why O forms a -2 charged ion. Provide a written explanation and Bohr-Rutherford diagrams for both the atom and the ion.
3. Write formulas for each of the following compounds.

sulfur pentafluoride
calcium oxide
nitrogen gas
iron (III) sulfide

sodium sulfate
tin(II) bromide
calcium phosphate
ammonia

4. Name the type of reaction and balance each equation.



5. Write word equations for each of the reactions in #4. C_3H_8 is propane.
6. Describe a standard test for each of the following substances:
 - a) oxygen gas
 - b) hydrogen gas
 - c) carbon dioxide gas
 - d) water vapour

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Clemens Wendtner

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