

Chemistry Review for Final

- Classify each of the following as homogeneous or heterogeneous mixtures:
a) blood b) chocolate chip cookies c) dissolved "koolaid" d) homogenized milk e) bronze
- Classify the following changes as physical or chemical:
a) food spoils b) an icicle melts c) a nail rusts
d) oil is pumped out of a well e) an egg is fried f) salt dissolves in water
g) a window is broken
- A compound is analyzed and is determined to be made up of 5.88% hydrogen, and 94.12% oxygen by mass. What is its empirical formula?
- Given the following percent compositions and molecular weights, determine the molecular formula for each compound.
a) 81.7% carbon, 18.3% hydrogen, molecular mass = 44.1 g/mol
b) 37.8% carbon, 6.4% hydrogen, 55.8% chlorine, molecular mass = 127.0 g/mol
- Describe the different types of intermolecular forces.
- Predict the monatomic ions formed by the following elements:
a) hydrogen b) magnesium c) sulfur d) iodine e) aluminum
- Balance the following equations. If they need no coefficients, write "balanced."
a) $\text{CF}_4(\text{l}) \rightarrow \text{C}(\text{s}) + \text{F}_2(\text{g})$
b) $\text{H}_2\text{SO}_4(\text{aq}) + \text{KOH}(\text{aq}) \rightarrow \text{KH}_2\text{SO}_4(\text{aq}) + \text{H}_2\text{O}(\text{l})$
c) $\text{ZnCl}_2(\text{aq}) + \text{H}_2(\text{g}) \rightarrow \text{Zn}(\text{s}) + \text{HCl}(\text{aq})$
d) $\text{SO}_2(\text{g}) + \text{H}_2\text{O}(\text{l}) + \text{O}_2(\text{g}) \rightarrow \text{H}_2\text{SO}_4(\text{aq})$
e) $\text{Li}(\text{s}) + \text{H}_2\text{O}(\text{l}) \rightarrow \text{LiOH}(\text{aq}) + \text{H}_2(\text{g})$
f) $\text{Cu}_2\text{O}(\text{aq}) + \text{Cu}_2\text{S}(\text{aq}) \rightarrow \text{Cu}(\text{s}) + \text{SO}_2(\text{g})$
g) $\text{Na}_2\text{SO}_4(\text{aq}) + \text{BaCl}_2(\text{aq}) \rightarrow \text{BaSO}_4(\text{s}) + \text{NaCl}(\text{aq})$
h) $\text{CH}_3\text{OH}(\text{l}) + \text{O}_2(\text{g}) \rightarrow \text{CO}_2(\text{g}) + \text{H}_2\text{O}(\text{g})$
- Classify each of the above according to the 5 types of reactions taken in class. If no classification fits, write "other."
- Write the formula for each material correctly and then balance the equation. For each reaction, tell what type of reaction it is. For some reactions, you will need to determine the products.
 - sulfur trioxide and water combine to make sulfuric acid.
 - lead (II) nitrate and sodium iodide react to make lead iodide and sodium nitrate.
 - calcium fluoride and sulfuric acid make calcium sulfate and hydrogen fluoride (Hydrofluoric acid)
 - calcium carbonate will come apart when you heat it to leave calcium oxide and carbon dioxide.
 - propane burns (with oxygen)
 - aluminum sulfate and calcium hydroxide become aluminum hydroxide and calcium sulfate.
 - copper metal and silver nitrate react to form silver metal and copper (II) nitrate.
 - sodium metal and chlorine react to make sodium chloride.
- Describe the difference between ionic and molecular compounds. You may answer in point form.
- Classify each of the following as ionic or molecular, and give the correct chemical name for each:
 - NaI
 - FeO
 - K_2S
 - GaBr_3
 - S_4N_2
 - $\text{Zn}(\text{CH}_3\text{COO})_2$
 - SnF_4
 - $\text{LiCl} \cdot 4\text{H}_2\text{O}$
 - P_2O_5
- Give the correct chemical formula for each of the following:
 - hydrochloric acid
 - sodium hexafluoride

Chemistry Final Review

J Rink



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Organic Chemistry Jonathan Clayden, Nick Greeves, Stuart Warren, 2012-03-15 A first and second year undergraduate organic chemistry textbook specifically geared to British and European courses and those offered in better schools in North America this text emphasises throughout clarity and understanding **Chemistry and Industry Review**, 1929

Organic Chemistry Mr. Rohit Manglik, 2024-05-07 Focuses on structure synthesis mechanisms and reactions of organic compounds **Pharmaceutical Organic Chemistry I - (Theory)** Mr. Rohit Manglik, 2024-07-24 In this book we will study about pharmaceutical organic chemistry i theory to understand its practical applications and theoretical foundations in the field of pharmacy and healthcare **DAT Prep Plus 2023-2024** Kaplan Test Prep, 2023-02-07 Kaplan s DAT Prep Plus 2023 2024 provides the test taking strategies realistic practice and expert guidance you need to score higher on the Dental Admissions Test Our comprehensive subject review reflects recent changes to the blueprint of the exam question types and test interface You ll get two full length practice DATs and expert tips to help you face Test Day with confidence We re so confident that DAT Prep Plus offers all the knowledge you need to excel at the DAT that we guarantee it after studying with our online resources and book you ll score higher on the DAT or you ll get your money back The Best Review Two updated full length online practice exams for test like practice Study planning guidance More than 600 practice questions for every subject with detailed answers and explanations 12 page full color study sheets for high yield review on the go A guide to the current DAT Blueprint so you know exactly what to expect on Test Day Comprehensive review of all of the content covered on the DAT Expert Guidance Our books and practice questions are written by veteran teachers who know students every explanation is written to help you learn Kaplan s experts ensure our practice questions and study materials are true to the test We invented test prep Kaplan www kaptest com has been helping students for 80 years and our proven strategies have helped legions of students achieve their dreams Publisher s Note Products purchased from 3rd party sellers are not guaranteed by the publisher for quality authenticity or access to any online entities included with the product **Idea Industry**, 2008 **FDA's Generic Drug Enforcement and Approval Process** United States. Congress. House. Committee on Energy and Commerce. Subcommittee on Oversight and Investigations, 1992 **Optimization of Solid-Phase Combinatorial Synthesis** Bing Yan, Anthony W. Czarnik, 2001-12-04 Addresses the key topic in combinatorial synthesis how to optimize the quality of a combinatorial library by determining the usefulness of synthesized compounds the reliability of biological assay results and analyzing academic and industrial applications real world examples and case studies of successful and unsuccessful technologies **The Normal News**, 1895 **Energy Research Abstracts**, 1989 **TRAC: Trends in Analytical Chemistry** U A Th Brinkman, J R Durig, P. Van Espen, 2013-09-24 TRAC Trends in Analytical Chemistry Volume 8 provides information pertinent to the trends in the field of analytical chemistry This book presents a variety of topics related to analytical chemistry including protein purification biotechnology Raman spectroscopy in pharmaceutical field

electrokinetic chromatography and flow injection analysis Organized into 50 chapters this volume begins with an overview of scientometric investigations that enable the quantitative study of the evolution of its various components and can thereby uncover how information is utilized to diffuse and generate knowledge This text then discusses the economic significance of sensing and control as being the main factors in determining process economics and in offering products and business opportunities Other chapters consider the important relationship between Raman spectroscopy and other analytical methods This book discusses as well the interfaces between a gas chromatograph and a Fourier transform infrared spectrometer The final chapter deals with chemometrics routines This book is a valuable resource for analytical chemists and biochemists

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United States. President's Biomedical Research Panel,1976 **Nobel Prizes And Nature's Surprises** Erling
Norrby,2013-10-10 Each year the Nobel Prizes in the natural sciences reveal amazing discoveries New milestones in the
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archives where nominations are kept secret for 50 years after the awards have been made They represent a treasure for real
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until 1959 The present book takes us up to 1962 surveying a range of dazzling discoveries All prizes in immunology are
reviewed Their impact on our capacity to control infectious diseases and transplant organs are highlighted The Nobel year
1962 is exceptional in recognizing the most major advance in biology since Darwin in 1859 presented his theory of evolution
This was the dramatic discovery of the double helix structure of DNA by Watson and Crick in 1953 The era of molecular
biology had begun Its explosive development continues into the present *Resources in Education* ,1993-11 *Basic Issues*
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