Name: \_\_\_\_\_\_ Class: \_\_\_\_\_ Date: \_\_\_\_\_ ID: A

#### CP Chemistry - Zajac - Semester II Exam Study Guide / Practice Exam

#### Multiple Choice

Identify the choice that best completes the statement or answers the question.

- - a. the number of atoms of bromine in 1 mol Be,
  - b. the number of atoms of gold in 1 mol Au-
  - c. the number of molecules of nitrogen in 1 mol N2
  - d. the number of molecules of carbon mesonade in 1 mol CO
- How many moles of tangeten atoms are in 4.8 × 10<sup>70</sup> atoms of tangeten?
  - $a_c = 8.0 \times 10^2$  moles.

c.  $1.3 \times 10^{-1}$  modes

8.0 × 10<sup>3</sup> moles.

- 1.5 × 10<sup>-1</sup> modes.
- Which of the following is NOT a true about atomic mass?
  - a. The atomic mass is 12 g for magnesium.
  - The atomic mass is the mass of one mole of atoms.
  - c. The atomic mass is found by checking the periodic table.
  - The atomic mass is the number of grants of an element that is numerically equal to the mass in amu.
- 4. The mass of a mole of NaCl is the
  - a. molar mass

c. molecular mess-

b. atomic mass

d. gram atomic mass

- What is the molar mass of AuCL?
  - a. 96 a.

e. 232.5 g

b. 130 g

- d. 300.6 g
- The molar mass of C<sub>1</sub>H<sub>10</sub>, and the molar mass of CaCO<sub>1</sub> contain approximately the same number of \_
  - a. carbon atoms

at applicable

by dentiones

- d. gerantes
- 7. How many moles of Calle, are in 5.0 grams of Calle,?
  - a. 2.5 × 10<sup>-7</sup> mod

4.0 × 80° mol

b. 4.2 × 10<sup>-2</sup> good

- d. 1.0 × 10 mol.
- The lowest whole-number ratio of the elements in a compound is called the \_\_\_\_\_\_
  - a: empirical formula

c. bistory formula

b. molecular formula

d. representative formula

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**EW Minium** 

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