

CHAPTER 4 PRACTICE TEST: SOLUTION STOICHIOMETRY1) Which of the following are weak electrolytes? HCl, $\text{HC}_2\text{H}_3\text{O}_2$, NH_3 , KCl

A) HCl, KCl

D) $\text{HC}_2\text{H}_3\text{O}_2$, NH_3 B) HCl, $\text{HC}_2\text{H}_3\text{O}_2$, NH_3 , KClE) HCl, $\text{HC}_2\text{H}_3\text{O}_2$, KClC) $\text{HC}_2\text{H}_3\text{O}_2$, KCl

Answer: D Diff: 2

2) What are the spectator ions in the reaction between KOH (aq) and HNO_3 (aq)?A) K^+ and H^+ C) K^+ and NO_3^- E) OH^- onlyB) H^+ and OH^- D) H^+ and NO_3^-

Answer: C Diff: 2

3) The balanced net ionic equation for precipitation of CaCO_3 when aqueous solutions of Na_2CO_3 and CaCl_2 are mixed is _____.A) $2\text{Na}^+(\text{aq}) + \text{CO}_3^{2-}(\text{aq}) \rightarrow \text{Na}_2\text{CO}_3(\text{aq})$

E)

B) $2\text{Na}^+(\text{aq}) + 2\text{Cl}^-(\text{aq}) \rightarrow 2\text{NaCl}(\text{aq})$ $\text{Na}_2\text{CO}_3(\text{aq}) + \text{CaCl}_2(\text{aq}) \rightarrow 2\text{NaCl}(\text{aq}) + \text{CaCO}_3(\text{s})$ C) $\text{Na}^+(\text{aq}) + \text{Cl}^-(\text{aq}) \rightarrow \text{NaCl}(\text{aq})$ D) $\text{Ca}^{2+}(\text{aq}) + \text{CO}_3^{2-}(\text{aq}) \rightarrow \text{CaCO}_3(\text{s})$

Answer: D Diff: 3

4) Which of the following are strong acids? HI, HNO_3 , HF, HBr

A) HF, HBr

C) HI, HF, HBr

E) HI, HNO_3 , HBrB) HI, HNO_3 , HF, HBrD) HNO_3 , HF, HBr

Answer: E Diff: 2

5) A neutralization reaction between an acid and a metal hydroxide produces _____.

A) water and a salt

C) oxygen gas

E) ammonia

B) hydrogen gas

D) sodium hydroxide

Answer: A Diff: 2

6) Which of these metals is the least easily oxidized? Na, Au, Fe, Ca, Ag

A) Na

B) Au

C) Fe

D) Ca

E) Ag

Answer: B Diff: 3

7) Based on the equations below, which metal is the most active?



A) Ni

B) Ag

C) Cu

D) Pb

E) N

Answer: A Diff: 3

8) What is the concentration (M) of KCl in a solution made by mixing 25.0 mL of 0.100 M KCl with 50.0 mL of 0.100 M KCl?

A) 0.100

B) 0.0500

C) 0.0333

D) 0.0250

E) 125

Answer: A Diff: 3

Chemistry Ch 18 Chemical Answer Key

John Anderson

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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Potenza, Loretta Jones, Lynn Koplitz, Thomas Spence, 2004-03-19 This combination manual is designed to help students avoid common mistakes and understand the material better The solutions manual section includes detailed answers and explanations to the odd numbered exercises in the text *NEET 2018 Chemistry Guide - 5th Edition* Disha

Experts, 2017-08-29 The thoroughly revised updated 5th Edition of NEET 2018 Chemistry Must for AIIMS JIPMER is developed on the objective pattern following the chapter plan as per the NCERT books of class 11 and 12 The new edition is empowered with an additional exercise which contains Exemplar past 5 year NEET 2013 2017 questions Concept Maps have been added for each chapter The book contains 31 chapters in all as per the NCERT books Each chapter provides exhaustive theory followed by a set of 2 exercises for practice The first exercise is a basic exercise whereas the second exercise is advanced The solutions to all the questions have been provided immediately at the end of each chapter The complete book has been aligned as per the chapter flow of NCERT class 11 12 books **Chemical Solution Deposition of Functional**

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photoelectrocatalytic solar cells In the appendix detailed cooking recipes for selected material systems are offered

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Charles Olsen,Alfred Melhado,1907 The issues for 1907 and 1909 contain a Review of chemical literature 43 *JEE Main*

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in the field thus preventing dyestuff manufacturing from becoming what might otherwise be viewed by now as a sunset

industry The volumes which followed have provided ample evidence for our belief that the field of colour chemistry is very

much alive though arguably in need of further stimulus For instance a viable approach to the design of new chromophores

and to the design of metal free acid direct and reactive dyes having fastness properties comparable to their metallized

counterparts represent the kind of breakthroughs that would help ensure the continued success of this important field While

it must be acknowledged that serendipity smiled on our discipline at its inception and has repeated the favor from time to time since then few would argue against the proposition that most of the significant advances in the technology associated with any scientific discipline result from research designed to enhance our understanding of the fundamental causes for experimental observations many of which are pursued because they are unexpected intriguing and intellectually stimulating Little reflection is required for one who knows the history of the dyestuff industry to realize that this is certainly true in the colour chemistry arena as it was basic research that led to fiber reactive dyes dyes for high technology and modern synthetic organic pigments

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