

ATOMIC STRUCTURES**MULTIPLE CHOICE QUESTIONS**

1. The wave function ψ of 2s – orbital is given by

$$\psi_{2s} = \frac{1}{2\sqrt{2\pi}} \left(\frac{1}{a_0} \right)^{3/2} \left(2 - \frac{r}{a_0} \right) e^{-r/2a_0}$$

At $r = r_0$, radial node is formed. Which of the following is correct?

- (a) $r_0 = 2a_0$ (b) $2r_0 = a_0$
(c) $r_0 = 3a_0$ (d) None of these is correct

Explanation: (a)

At $r = r_0$, $\psi_{2s}^2 = 0$

It is possible only when $2 - \frac{r}{a_0} = 0$

$\therefore r_0 = 2a_0$

2. Which of the following set of quantum numbers belong to highest energy?

- (a) $n = 2, l = 1, m = 0, s = +\frac{1}{2}$
(b) $n = 3, l = 0, m = 0, s = +\frac{1}{2}$
(c) $n = 4, l = 0, m = 1, s = +\frac{1}{2}$
(d) $n = 4, l = 1, m = 1, s = +\frac{1}{2}$

Explanation: (d)

Chemistry Atomic Theory Questions

RJ Shavelson



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