

## SNC2DN -CHEMISTRY Diagnostic Test

Name: \_\_\_\_\_ Date: \_\_\_\_\_

**NOTE:** The purpose of this test is to illustrate your current knowledge about specific domains in science. Your results on this test will in no way directly affect course grade.

**INSTRUCTIONS:** Be as complete as possible. Do not go over the space provided. Fit the most important things that you have to say in the space provided. Please write in complete sentences.

### CHEMISTRY

1. What is matter?
2. What is the difference between an element, an atom, and a compound?
3. What happens in a chemical reaction?
4. Distinguish between a physical and chemical property. Provide examples of each.
5. Label each of the following properties as either physical or chemical:
  - a. Copper sulfate crystals are blue. \_\_\_\_\_
  - b. Gold is an excellent conductor of electricity. \_\_\_\_\_
  - c. Iron rusts when exposed to air and water. \_\_\_\_\_
  - d. Salt is soluble in water. \_\_\_\_\_
  - e. Magnesium reacts with acid to produce hydrogen gas. \_\_\_\_\_
  - f. Gasoline burns in a car engine. \_\_\_\_\_
6. List 5 clues that indicate a chemical change has occurred.
7. (a) What is the formula for density?  
  
(b) Calculate the density of a 500g rectangular block with the dimensions 8cm x 6cm x 5cm.
8. Define the terms: ion, anion and cation.
9. What 3 main categories of elements make up the periodic table?
10. Define the terms periods and families/groups as they relate to the Periodic table.

# Chemistry Answer Key Diagnostic Test Topic

**M Carnoy**



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