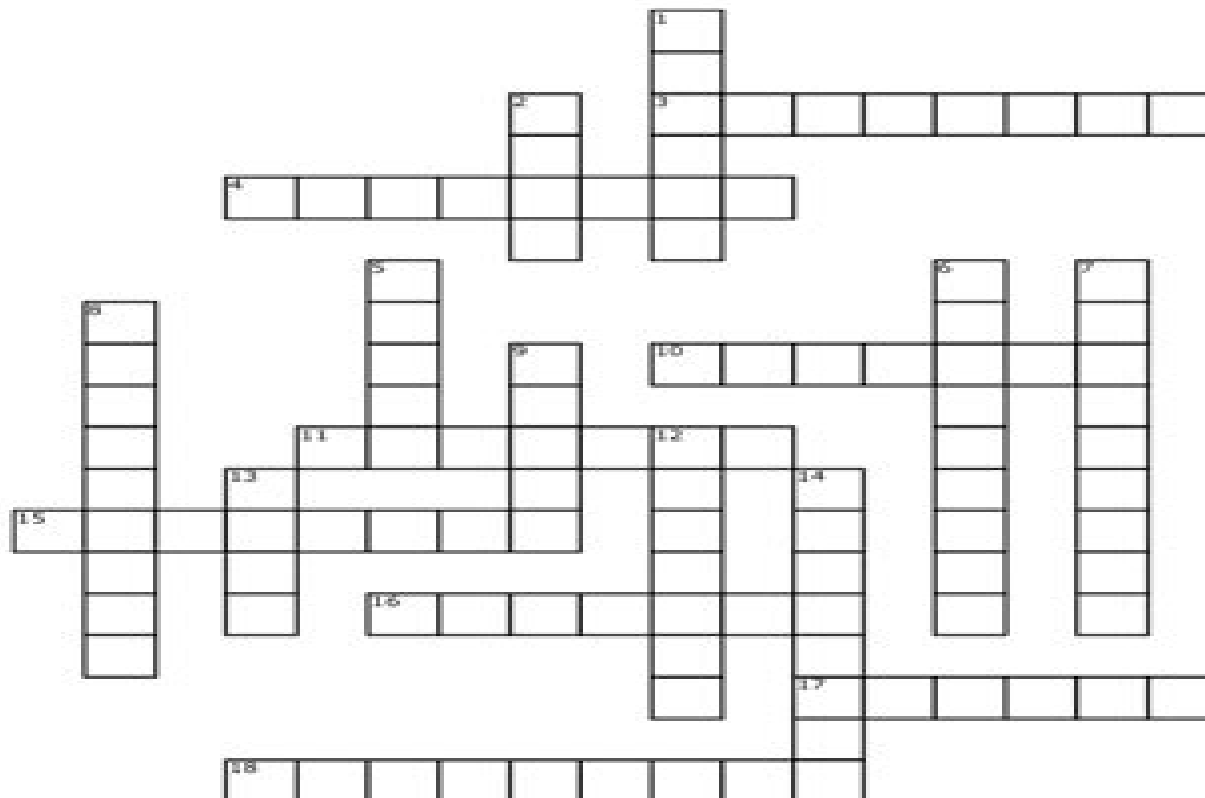


Water Chemistry



Across

3. A _____ is a combination of atoms formed by a covalent bond.
 4. The two _____ share their electrons.
 10. Eight electrons fill the outer most shell or _____.
 11. _____ have a positive charge.
 15. The bond between two atoms of hydrogen is called a _____ bond.
 16. The term _____ is applied to any substance that cannot be decomposed into simpler substances by chemical reactions.
 17. O is the symbol for _____.

18. _____ have a negative charge.

Down

1. The atomic _____ is determined by the number of protons.
 2. An _____ is the smallest part of any element that retains all the properties of that element.
 5. Water is a _____ molecule.
 6. Chemists have identified _____ chemical elements in nature.
 7. Each element is comprised of _____ particles called atoms.
 8. Each atom is made up of 3 types of _____ particles.

9. The valence's eight electrons can also be called an _____.

12. The atom's _____ has an overall positive charge equal in magnitude to the number of protons it contains.
 13. Atomic _____ is determined by the sum of protons and neutrons.
 14. _____ have no charge. They are neutral.

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Yan Bai



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