Your name (write clearly):

General Chemistry, CHEM 101

Exam 1, 02/25/2021 @08:00 AM

Exam duration is 40 minutes

The exam needs to be uploaded to Canvas in one single file before 08:50 am

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Problem 1. (30 points)

PART 1. Write the FULL IONIC equation for each of the following reactions? (20 pnts)

A) $K_2SO_4(ag) + 2 AgNO_3(ag) \rightarrow 2 KNO_3(ag) + Ag_2SO_4(s)$

$$2K^{*}(aq) + SO_{4}^{2-}(aq) + 2 Ag^{*}(aq) + 2 NO_{3}^{-}(aq)$$

 $\rightarrow 2K^{*}(aq) + 2 NO_{3}^{-}(aq) + Ag_{2}SO_{4}(s)$

B) Na₂CO₂(ag) + 2 HCl(ag) → 2 NaCl(ag) + CO₂(g) + H₂O(I)

$$2 \text{ Ma}^{+}(aq) + \text{CO}_{3}^{2-}(aq) + 2 \text{ H}^{+}(aq) + 2 \text{ GA}^{-}(aq) \rightarrow 2 \text{ Ma}^{+}(aq) + 2 \text{ GA}^{-}(aq) + \text{CO}_{2}(g) + \text{H}_{2}\text{O}(l)$$

PART 2. Write the NET IONIC equations of the previous two reactions? (10 pnts)

A()

$$2 \text{ Ag}^+(aq) + \text{SO}_4^{2-}(aq) \rightarrow \text{Ag}_2 \text{SO}_4(s)$$

B):

$$CO_3^{2-}(aq) + 2 H^+(aq) \rightarrow CO_2(g) + H_2O(l)$$

Problem 2. (30 points)

PART 1. Consider the following octane (CsH₁₈) combustion reaction:

$$2 \text{ CsH}_{18}(I) + 25 \text{ O}_2(g) \rightarrow 16 \text{ CO}_2(g) + 18 \text{ H}_2\text{O}(g)$$

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Table of Contents Chemistry 101 Placement Test Study Guide Csun

- 1. Understanding the eBook Chemistry 101 Placement Test Study Guide Csun
 - The Rise of Digital Reading Chemistry 101 Placement Test Study Guide Csun
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Chemistry 101 Placement Test Study Guide Csun
 - Exploring Different Genres
 - o Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemistry 101 Placement Test Study Guide Csun
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Chemistry 101 Placement Test Study Guide Csun
 - Personalized Recommendations
 - Chemistry 101 Placement Test Study Guide Csun User Reviews and Ratings
 - Chemistry 101 Placement Test Study Guide Csun and Bestseller Lists

- 5. Accessing Chemistry 101 Placement Test Study Guide Csun Free and Paid eBooks
 - Chemistry 101 Placement Test Study Guide Csun Public Domain eBooks
 - Chemistry 101 Placement Test Study Guide Csun eBook Subscription Services
 - Chemistry 101 Placement Test Study Guide Csun Budget-Friendly Options
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 - Chemistry 101 Placement Test Study Guide Csun Compatibility with Devices
 - Chemistry 101 Placement Test Study Guide Csun Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chemistry 101 Placement Test Study Guide Csun
 - Highlighting and Note-Taking Chemistry 101 Placement Test Study Guide Csun
 - Interactive Elements Chemistry 101 Placement Test Study Guide Csun
- 8. Staying Engaged with Chemistry 101 Placement Test Study Guide Csun
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemistry 101 Placement Test Study Guide Csun
- 9. Balancing eBooks and Physical Books Chemistry 101 Placement Test Study Guide Csun
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemistry 101 Placement Test Study Guide Csun
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Chemistry 101 Placement Test Study Guide Csun
 - Setting Reading Goals Chemistry 101 Placement Test Study Guide Csun
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chemistry 101 Placement Test Study Guide Csun
 - Fact-Checking eBook Content of Chemistry 101 Placement Test Study Guide Csun
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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