

**CHAPTER 7 REVIEW***Chemical Formulas and Chemical Compounds***SECTION 7-1****SHORT ANSWER** Answer the following questions in the space provided.

1. C In a Stock name such as iron(III) sulfate, the roman numeral tells us \_\_\_\_\_  
(a) how many atoms of Fe are in one formula unit  
(b) how many sulfate ions can be attached to the iron atom  
(c) the charge on each Fe ion  
(d) the total positive charge of the formula unit
2. C The result of changing a subscript in a correctly written chemical formula is to \_\_\_\_\_  
(a) change the number of moles represented by the formula  
(b) change the charges on the other ions in the compound  
(c) change the formula so that it no longer represents the compound it previously represented  
(d) have no effect on the formula
3. The explosive TNT has the molecular formula  $C_7H_5(NO_2)_3$ .  
4 elements a. How many elements make up this compound?  
6 oxygen atoms b. How many oxygen atoms are present in one molecule of  $C_7H_5(NO_2)_3$ ?  
21 atoms c. How many atoms in total are present in one molecule of  $C_7H_5(NO_2)_3$ ?  
 $4.2 \times 10^{24}$  atoms d. How many atoms are present in a sample of  $2 \times 10^{23}$  molecules of  $C_7H_5(NO_2)_3$ ?
4. How many atoms are present in each of these formula units?  
11 atoms a.  $Ca(HCO_3)_2$   
45 atoms b.  $C_{12}H_{22}O_{11}$   
10 atoms c.  $Fe(ClO_2)_3$   
9 atoms d.  $Fe(ClO_3)_2$
5.  $N_2O_5$  a. What is the formula for the compound dinitrogen pentoxide?  
carbon(IV) sulfide b. What is the Stock name for the covalent compound  $CS_2$ ?  
 $H_2SO_3$  c. What is the formula for sulfurous acid?  
phosphoric acid d. What is the name for the acid  $H_3PO_4$ ?

# Chemical Formulas And Chemical Compounds Study Guide

**YCT Expert Team**



## Chemical Formulas And Chemical Compounds Study Guide:

**Chemistry, Student Study Guide** James E. Brady, Fred Senese, 2008-01-28 The image on the front cover depicts a carbon nanotube emerging from a glowing plasma of hydrogen and carbon as it forms around particles of a metal catalyst Carbon nanotubes are a recently discovered allotrope of carbon Three other allotropes of carbon buckyballs graphite and diamond are illustrated at the left as is the molecule methane  $\text{CH}_4$  from which nanotubes and buckyballs can be made The element carbon forms an amazing number of compounds with structures that follow from simple methane found in natural gas to the complex macromolecules that serve as the basis of life on our planet The study of chemistry also follows from the simple to the more complex and the strength of this text is that it enables students with varied backgrounds to proceed together to significant levels of achievement Barron's Science 360: A Complete Study Guide to Chemistry with Online Practice Barron's Educational Series, Mark Kernion, Joseph A. Mascetta, 2021-09-07 provides a complete guide to the fundamentals of chemistry Page 4 of cover The Practice of Chemistry Study Guide & Solutions Manual Pamela Mills, Amina El-Ashmawy, 2003-04-14 Designed to help students understand the material better and avoid common mistakes Also includes solutions and explanations to odd numbered exercises *Hazmat Chemistry Study Guide (Second Edition)* Jill Meryl Levy, 2011 Student Study Guide/Solutions Manual for Essentials of General, Organic, and Biochemistry Denise Guinn, Rebecca Brewer, Rachel C. Lum, 2009-09-15 The Student Study Guide and Solutions Manual provides students with a combined manual designed to help them avoid common mistakes and understand key concepts After a brief review of each section's critical ideas students are taken through stepped out worked examples try it yourself examples and chapter quizzes all structured to reinforce chapter objectives and build problem solving techniques The solutions manual includes detailed solutions to all odd numbered exercises in the text *Basic Concepts of Chemistry, 9e Study Guide and Solutions Manual* Leo J. Malone, Theodore O. Dolter, 2012-01-03 The 9th edition of Malone's Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections assessment exercises at the end each section and relevant chapter problems at the end of each chapter A new Math Check allows quick access to the needed basic skill The first chapter now includes brief introductions to several fundamental chemical concepts and Chapter Synthesis Problems have been added to the end of each chapter to bring key concepts into one encompassing problem Every concept in the text is clearly illustrated with one or more step by step examples Making it Real essays have been updated to present timely and engaging real world applications emphasizing the relevance of the material they are learning This edition continues the end of chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter **ASVAB Study Guide Premium: 6 Practice Tests + Comprehensive Review + Online Practice** Barron's Educational

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