

Name key

## Physical Science Chemical Bonding Study Guide

### A. Covalent & Ionic Compounds/Bonds

1. When an atom loses an electron it becomes a positive ion (cation), and when it gains an electron it becomes a negative ion (anion).
2. An element in the halogen group has 7 valence electrons, and wants to gain 1 more to have 8. This is because of the octet rule.
3. An ionic bond is the attraction between oppositely charged ions, AKA a metal and a nonmetal.
4. In a covalent bond, atoms share electrons, whereas in an ionic bond, atoms transfer electrons.
5. (Honors only) When electrons are shared unequally in a covalent bond it is polar, and when they're shared equally it is nonpolar. You can predict if a compound is ionic, polar covalent or nonpolar covalent by subtracting the electronegativities.
6. An atom can have up to 8 valence electrons. The family (column) on the periodic table that all have 8 valence electrons is called noble gases.
7. Atoms can gain, lose, or share valence electrons to become more stable.

### B. Formulas for Ionic compounds - Fill in the chart below

	Magnesium	Fluorine	Nitrogen	Lithium
# OF VALENCE ELECTRONS	<u>2</u>	<u>7</u>	<u>5</u>	<u>1</u>
OXIDATION #	<u>+2</u>	<u>-1</u>	<u>-3</u>	<u>+1</u>

2. What is the **chemical formula** for each of the following ionic compounds?  
(*hint: you can either draw the Lewis structures and use arrows to show the transfer of electrons, OR use the periodic table to find their oxidation numbers. They will combine so that their oxidation numbers cancel out to zero. Use the criss cross method if their oxidation numbers do not cancel.*)

- a) Potassium and Chlorine - KCl
- b) Calcium and Bromine - CaBr<sub>2</sub>
- c) Magnesium and oxygen - MgO

# Chemical Bond Study Guide For Freshman

**David Kirk**



## **Chemical Bond Study Guide For Freshman:**

## Unveiling the Energy of Verbal Beauty: An Psychological Sojourn through **Chemical Bond Study Guide For Freshman**

In a world inundated with monitors and the cacophony of quick interaction, the profound energy and mental resonance of verbal art usually disappear into obscurity, eclipsed by the continuous onslaught of noise and distractions. However, nestled within the lyrical pages of **Chemical Bond Study Guide For Freshman**, a fascinating perform of fictional elegance that pulses with raw feelings, lies an memorable journey waiting to be embarked upon. Composed by way of a virtuoso wordsmith, that mesmerizing opus manuals readers on a psychological odyssey, delicately revealing the latent potential and profound affect stuck within the intricate web of language. Within the heart-wrenching expanse with this evocative examination, we shall embark upon an introspective exploration of the book is central styles, dissect their captivating writing type, and immerse ourselves in the indelible impact it leaves upon the depths of readers souls.

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### **Table of Contents Chemical Bond Study Guide For Freshman**

1. Understanding the eBook Chemical Bond Study Guide For Freshman
  - The Rise of Digital Reading Chemical Bond Study Guide For Freshman
  - Advantages of eBooks Over Traditional Books
2. Identifying Chemical Bond Study Guide For Freshman
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Chemical Bond Study Guide For Freshman
  - User-Friendly Interface
4. Exploring eBook Recommendations from Chemical Bond Study Guide For Freshman

- Personalized Recommendations
  - Chemical Bond Study Guide For Freshman User Reviews and Ratings
  - Chemical Bond Study Guide For Freshman and Bestseller Lists
5. Accessing Chemical Bond Study Guide For Freshman Free and Paid eBooks
    - Chemical Bond Study Guide For Freshman Public Domain eBooks
    - Chemical Bond Study Guide For Freshman eBook Subscription Services
    - Chemical Bond Study Guide For Freshman Budget-Friendly Options
  6. Navigating Chemical Bond Study Guide For Freshman eBook Formats
    - ePub, PDF, MOBI, and More
    - Chemical Bond Study Guide For Freshman Compatibility with Devices
    - Chemical Bond Study Guide For Freshman Enhanced eBook Features
  7. Enhancing Your Reading Experience
    - Adjustable Fonts and Text Sizes of Chemical Bond Study Guide For Freshman
    - Highlighting and Note-Taking Chemical Bond Study Guide For Freshman
    - Interactive Elements Chemical Bond Study Guide For Freshman
  8. Staying Engaged with Chemical Bond Study Guide For Freshman
    - Joining Online Reading Communities
    - Participating in Virtual Book Clubs
    - Following Authors and Publishers Chemical Bond Study Guide For Freshman
  9. Balancing eBooks and Physical Books Chemical Bond Study Guide For Freshman
    - Benefits of a Digital Library
    - Creating a Diverse Reading Collection Chemical Bond Study Guide For Freshman
  10. Overcoming Reading Challenges
    - Dealing with Digital Eye Strain
    - Minimizing Distractions
    - Managing Screen Time
  11. Cultivating a Reading Routine Chemical Bond Study Guide For Freshman
    - Setting Reading Goals Chemical Bond Study Guide For Freshman
    - Carving Out Dedicated Reading Time
  12. Sourcing Reliable Information of Chemical Bond Study Guide For Freshman

- Fact-Checking eBook Content of Chemical Bond Study Guide For Freshman
- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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