

Chemical Equations and Reactions

SECTION 2

SHORT ANSWER Answer the following questions in the space provided.

1. Match the equation type on the left to its representation on the right.

___ **c** ___ synthesis



___ **d** ___ decomposition



___ **b** ___ single-displacement



___ **a** ___ double-displacement



2. ___ **c** ___ In the reaction described by the equation $2Al(s) + 3Fe(NO_3)_2(aq) \rightarrow 3Fe(s) + 2Al(NO_3)_3(aq)$, iron has been replaced by

(a) nitrate.

(c) **aluminum.**

(b) water.

(d) nitrogen.

3. ___ **a** ___ Of the following chemical equations, the only reaction that is both synthesis and combustion is



4. ___ **b** ___ Of the following chemical equations, the only reaction that is both combustion and decomposition is



(d)

5. Identify the products when the following substances decompose:

a. **Its separate elements** _____ a binary compound

b. **Metal oxide + water** _____ most metal hydroxides

c. **Metal oxide + carbon dioxide** _____ a metal carbonate

d. **Water + sulfur dioxide** _____ the acid H_2SO_3

6. The complete combustion of a hydrocarbon in excess oxygen yields the products

Chemical Reactions Chapter 8 Exploring Physical Science

Pasquale De Marco



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Matter, Building Block of the Universe, 1993

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