

# Chemical Reactions and Equations

10

## A. Magnesium and Oxygen

1. Initial appearance of Mg      Shiny light gray ribbon of metal
2. Evidence of chemical reaction Heat and light energy produced  
White ash produced
3. Balance:  $2 \text{Mg}_{(s)} + \text{O}_{2(g)} \xrightarrow{\Delta} 2 \text{MgO}_{(s)}$
4. Type of chemical reaction: Combination reaction

## B. Zinc and Copper(II) Sulfate

Time	$\text{CuSO}_4(aq)$ Appearance	$\text{Zn}(s)$ Appearance	Evidence of a Chemical Reaction
1. initial	Blue/green	Dull gray	
2. after 30 min	Lt blue	Black	Change in color

3.  $\text{Zn}_{(s)} + \text{CuSO}_{4(aq)} \xrightarrow{\Delta} \text{Cu}_{(s)} + \text{ZnSO}_{4(aq)}$  (the equation is balanced)

4. Type of chemical reaction: Single replacement reaction

## C. Reactions of Metals and HCl

Metal	1. Appearance of Metals	2. Evidence of a Chemical Reaction
$\text{Zn}(s)$	Gray, shiny pebbles	Formation of bubbling
$\text{Cu}(s)$	Orange, shiny strips	No change
$\text{Mg}(s)$	Light gray, shiny strips	Formation of many bubbles Heat produced

# Chemical Reactions And Equations Lab Answers

**S Ashworth**



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