

Challenging Calculations - Study-Guide

4. How many molecules are there in 2.48 moles of CO_2 ?

It is 2.48 x 10^{23} molecules

5. How many formula units and grams is 1.5 moles of CaBr_2 ?

It is 1.5 x 10^{23} fu

6. How many molecules are there in 8 moles of hydrogen azide, HN_3 ?

It is molecules

7. How many formula units of oxygen is 1

It is 10²³ fu

8. How many formula units of oxygen is 2

It is 2 x 10²³ fu

9. What is the mass formula units of CaCl_2 ?

It is 119.1 g/mol

10. What is the number of moles in 1.5 grams of CaCl_2 ?

It is 0.0126 moles

11. How many formula units of H_2O are there in 100 g. What is the number of moles of H_2O ?

It is 5.56 x 10²⁴ fu and 5.56 x 10²⁴ fu

It is 5.56 x 10²⁴ fu

12. If 100 g of H_2O is present, how many moles of H_2O are there? What is the mass of H_2O ?

It is 5.56 x 10²⁴ fu

13. What is the mass of H_2O in 100 g of H_2O ?

It is 100 g

14. What is the mass of H_2O in 100 g of H_2O ?

It is 100 g

Chemical Quantities Study Guide Answers

S Ben Porath



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