

Naming Ionic Compounds – Answer Key

Give the name of the following ionic compounds:

| | | Name |
|-----|-------------------------------|-------------------------------|
| 1) | Na_2CO_3 | sodium carbonate |
| 2) | NaOH | sodium hydroxide |
| 3) | MgBr_2 | magnesium bromide |
| 4) | KCl | potassium chloride |
| 5) | FeCl_2 | iron (II) chloride |
| 6) | FeCl_3 | iron (III) chloride |
| 7) | Zn(OH)_2 | zinc hydroxide |
| 8) | Be_2SO_4 | beryllium sulfate |
| 9) | CrF_2 | chromium (II) fluoride |
| 10) | Al_2S_3 | aluminum sulfide |
| 11) | PbO | lead (II) oxide |
| 12) | Li_3PO_4 | lithium phosphate |
| 13) | TiI_4 | titanium (IV) iodide |
| 14) | Co_3N_2 | cobalt (II) nitride |
| 15) | Mg_3P_2 | magnesium phosphide |
| 16) | $\text{Ga(NO}_2)_3$ | gallium nitrite |
| 17) | Ag_2SO_3 | silver sulfite |
| 18) | NH_4OH | ammonium hydroxide |
| 19) | Al(CN)_3 | aluminum cyanide |
| 20) | $\text{Be(CH}_3\text{COO)}_2$ | beryllium acetate |

Chemfiesta Naming Compunds Practice Answer Key

Sabine Zange



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Table of Contents Chemfiesta Naming Compounds Practice Answer Key

1. Understanding the eBook Chemfiesta Naming Compounds Practice Answer Key
 - The Rise of Digital Reading Chemfiesta Naming Compounds Practice Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemfiesta Naming Compounds Practice Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemfiesta Naming Compounds Practice Answer Key
 - User-Friendly Interface

4. Exploring eBook Recommendations from Chemfiesta Naming Compunds Practice Answer Key
 - Personalized Recommendations
 - Chemfiesta Naming Compunds Practice Answer Key User Reviews and Ratings
 - Chemfiesta Naming Compunds Practice Answer Key and Bestseller Lists
5. Accessing Chemfiesta Naming Compunds Practice Answer Key Free and Paid eBooks
 - Chemfiesta Naming Compunds Practice Answer Key Public Domain eBooks
 - Chemfiesta Naming Compunds Practice Answer Key eBook Subscription Services
 - Chemfiesta Naming Compunds Practice Answer Key Budget-Friendly Options
6. Navigating Chemfiesta Naming Compunds Practice Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Chemfiesta Naming Compunds Practice Answer Key Compatibility with Devices
 - Chemfiesta Naming Compunds Practice Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chemfiesta Naming Compunds Practice Answer Key
 - Highlighting and Note-Taking Chemfiesta Naming Compunds Practice Answer Key
 - Interactive Elements Chemfiesta Naming Compunds Practice Answer Key
8. Staying Engaged with Chemfiesta Naming Compunds Practice Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemfiesta Naming Compunds Practice Answer Key
9. Balancing eBooks and Physical Books Chemfiesta Naming Compunds Practice Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemfiesta Naming Compunds Practice Answer Key
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Chemfiesta Naming Compunds Practice Answer Key
 - Setting Reading Goals Chemfiesta Naming Compunds Practice Answer Key
 - Carving Out Dedicated Reading Time

12. Sourcing Reliable Information of Chemfiesta Naming Compunds Practice Answer Key
 - Fact-Checking eBook Content of Chemfiesta Naming Compunds Practice Answer Key
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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