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Quiz #4 – Combustion Analysis Name: Shaili Batsri

CHEM 1411 – Please give Empirical & Molecular Formula!

1) A 0.450 g sample of an oily liquid substance containing C, H & O produces 1.023 g CO₂ and 0.419 g H₂O in a combustion analysis. What is the empirical formula?

Empirical Formula: C₃H₆O

2) The oily liquid smells like pineapple, is a key ingredient used as a flavor enhancer in processed orange juices, and has a molar mass is approximately 116 g/mol, what is the molecular formula?

Molecular Formula: C₆H₁₂O₂

3) Write a balanced chemical equation for the above substance.



Chem 1411 Lab Manual Experiment 10 Answers

M Woodhall



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