

Final Practice Exam

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1. If a sample of matter is uniform throughout and cannot be separated into other substances by physical means, it is _____.
 - a. a compound
 - b. either a compound or an element
 - c. a homogeneous mixture
 - d. a heterogeneous mixture
 - e. an element
2. What volume of 10.0 M H_2SO_4 is required to prepare 4.0 L of 0.50 M H_2SO_4 ?
 - a. 0.20 L
 - b. 0.40 L
 - c. 0.50 L
 - d. 1.0 L
 - e. 4.0 L
3. A solution of silver nitrate is mixed with a solution of potassium fluoride. If a precipitate forms, the precipitate is:
 - a. Silver fluoride
 - b. Potassium nitrate
 - c. Potassium fluorate
 - d. Nitric fluoride
 - e. No precipitate is formed
4. If 5.0 mol of both hydrochloric acid and sodium sulfide are mixed and reacted according to the equation below, how many moles of hydrogen sulfide (H_2S) are produced?
$$\text{HCl} + \text{Na}_2\text{S} \rightarrow \text{H}_2\text{S} + \text{NaCl}$$
 - a. 1 mol
 - b. 1.25 mol
 - c. 2.5 mol
 - d. 3 mol
 - e. 5 mol

Chem 105 Practice Final Exam

Frank Pierce Foster



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