

LABORATORY REPORT

Experiment 7 Properties and Reactivity of Aqueous Solutions

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Score: _____

I. Introduction

Ionic and molecular (covalent) chemicals are two separate groups of soluble substances. A metal that tends to lose electrons and become a cation forms an ionic bond with a non-metal element that tends to gain electrons and become an anion, forming an ionic compound. While molecules such as sugar and methanol may dissolve in water, they do not transmit electricity because the elements remain in their molecular state rather than dissolving into ions. Ionic substances, such as NaCl, ionize in water and split into positively and negatively charged ions that are electrically conductive. Ions are charged particles with electrical conductivity. The chemical will ionize in a solution to become an excellent electrical conductor that may be used to create an electric circuit, allowing us to watch the lightbulb glow. Strong acids and ionic compounds, for example, are examples of compounds that totally dissolve in water because they have a high ion-solvent interaction. Weak acids and bases are examples of substances whose ions do not entirely dissociate in water, making them weak electrolytes. Because they don't contain charged particles, non-electrolytes such as soluble organic molecules and insoluble ionic compounds don't carry electricity. (General Aqueous Solution Properties, 2022). Since the presence of ions in a solution indicates the formation of ions, electrical conductivity is seen as proof of a chemical reaction. This experiment allowed us to identify various aqueous compounds, compare the conductivity of strong, weak, and nonelectrolytes, and watch the chemical interactions that occurred between them. To show the electrical conductivity of the various solutions in this test, we employed a conductivity device with a lightbulb. To prevent inaccurate bulb observation when utilizing the conductivity apparatus, we must make careful to rinse the electrodes with DI water. When handling HCl, calcium nitrate, and ammonium oxide, we must exercise caution and wear protective goggles and gloves due to their corrosive nature and potential for eye harm.

II. Results Analysis CHECK DATA SHEET

A. Data CHECK DATA SHEET

B. Discussion

A. Reaction between 0.1 M HC₂H₃O₂ and 0.1 M NH₄OH

In this part, we first saw two distinct reactants, mixed them, and then observed them once again. Since ammonium hydroxide and acetic acid both partially ionize in water, as we learned in the previous section, acetic acid is a weak electrolyte. However, when they combine, they produce an acid-base reaction.



The fact that both light bulbs glowed when the group watched indicates that the fluid had solidified into an electrolyte. Water and ammonium acetate are the products of combining the two substances. Ammonium acetate is a powerful electrolyte because it dissolves entirely in

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Mass spectrometry is fast becoming an indispensable field for medical professionals. The mass spectrometric analysis of metabolites and proteins promises to revolutionize medical research and clinical diagnostics. As this technology rapidly enters the medical field, practicing professionals and students need to prepare to take full advantage of its capabilities. *Medical Applications of Mass Spectrometry* addresses the key issues in the medical applications of mass spectrometry at the level appropriate for the intended readership. It will go a long way to help the utilization of mass spectrometry in medicine. The book comprises five parts. A general overview is followed by a description of the basic sampling and separation methods in analytical chemistry. In the second part, a solid foundation in mass spectrometry and modern techniques of data analysis is presented. The third part explains how mass spectrometry is used in exploring various classes of biomolecules including proteins and lipids. In the fourth section, mass spectrometry is introduced as a diagnostic tool in clinical treatment, infectious pathogen research, neonatal diagnostics, cancer, brain and allergy research, as well as in various fields of medicine: cardiology, pulmonology, neurology, psychiatric diseases, hematology, oncology, urologic diseases, gastrointestinal diseases, gynecology and pediatrics. The fifth part covers emerging applications in biomarker discovery and in mass spectrometric imaging. Provides a broad look at how the medical field is benefiting from advances in mass spectrometry. Guides the reader from basic principles and methods to cutting edge applications. There is NO comparable book on the market to fill this fast growing field. *Chemical Abstracts* , 1907 Nanotechnology Applications for Clean Water Mamadou Diallo, Jeremiah Duncan, Nora Savage, Anita Street, Richard Sustich, 2009-02-12

The World Health Organization in 2004 estimated approximately 1.1 billion people did not have access to clean water and that 35% of Third World residents died from water borne illnesses. While the situation is grim, recent advances strongly indicate that many of the current water quality problems can be addressed and potentially resolved using nanotechnology. Nanotechnology is already having a dramatic impact on research in water quality and *Nanotechnology Applications for Clean Water* highlights both the challenges and the opportunities for nanotechnology to positively influence this area of environmental protection. Here you will find detailed information on breakthroughs, cutting edge technologies, current research and future trends that may affect acceptance of widespread applications. The first four parts of the book cover specific topics including using nanotechnology for clean drinking water in both large scale water treatment plants and in point of use systems. For instance, recent advances show that many of the current problems involving water quality can be

addressed using nanosorbents nanocatalysts bioactive nanoparticles nanostructured catalytic membranes and nanoparticle enhanced filtration The book also discusses existing technologies and future potential for groundwater remediation pollution prevention and sensors The final part discusses the inherent societal implications that may affect acceptance of widespread applications Over 80 leading experts from around the world share their wealth of knowledge in this truly unique reference Institutions such as Center for the Purification of Water and Systems Univ of Illinois at Urbana Champaign UCLA Water Technology Center Carnegie Mellon University University of Kentucky The University of Western Ontario Pacific Northwest National Laboratory National Institute for Advanced Industrial Science and Technology Japan Munasinghe Institute for Development Sri Lanka and the Woodrow Wilson Center for Scholars are just a few of the knowledge centers represented in this book Water quality is a serious global issue in which government bodies and scientific communities face many challenges in ensuring clean water is available to everyone Nanotechnology is already showing dramatic results and this book is an attempt to share current technologies and future possibilities in reaching this goal From the Foreword Researchers and practitioners may find in this volume key challenges regarding clean water resources The presentations may crystallize new research and education programs Mihail Roco U S National Science Foundation and U S Nanotechnology Initiative Contributors from the US India Canada Japan UK Sri Lanka and South Africa Provides detailed information on breakthroughs cutting edge technologies current research and future trends that may affect acceptance of widespread applications Covers specific topics including using nanotechnology for clean drinking water in both large scale water treatment plants and in point of use systems Discusses existing technologies and future potential for groundwater remediation pollution prevention and sensors Highlights both the challenges and the opportunities for nanotechnology to positively influence this area of environmental protection

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