



# CHEMISTRY

## CHEAT SHEET



# INTERMOLECULAR FORCES

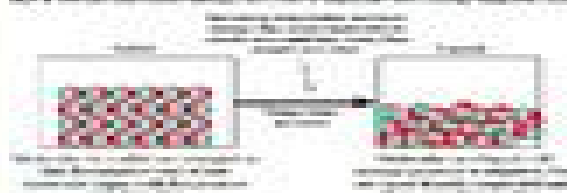
## INTERMOLECULAR FORCES CHEAT SHEET

Intermolecular forces are attractive forces between molecules. They are responsible for many physical properties and chemical behavior of substances.

The strength of intermolecular forces depends on the nature of the molecules. The order of intermolecular forces is as follows: Hydrogen bonding > Dipole-dipole interaction > London dispersion forces.

Intermolecular forces	Description	Types of molecules
Hydrogen bonding	Occurs when hydrogen is bonded to highly electronegative atoms (F, O, N). The resulting partial positive charge on hydrogen interacts with the partial negative charge on the electronegative atom.	Small molecules containing F, O, or N
Dipole-dipole interaction	Occurs between polar molecules. The positive end of one molecule is attracted to the negative end of another molecule.	Polar molecules
London dispersion forces	Occurs in all molecules, even non-polar ones. It is caused by temporary dipoles that arise from the movement of electrons.	All molecules

Intermolecular forces are responsible for many physical properties of substances, such as boiling point, melting point, and viscosity. The strength of intermolecular forces increases with the molecular weight and the number of polar groups in the molecule.



The order of intermolecular forces is as follows: Hydrogen bonding > Dipole-dipole interaction > London dispersion forces.

	Formula	Boiling Point	Melting Point
Water	$H_2O$	100°C	0°C
Alcohol	$C_2H_5OH$	78°C	-114°C
Gasoline	$C_8H_{18}$	125°C	-95°C

# Chemistry Physics Forces Study Guide

**DP Hallahan**



## **Chemistry Physics Forces Study Guide:**

## **Chemistry Physics Forces Study Guide** Book Review: Unveiling the Power of Words

In a world driven by information and connectivity, the ability of words has become more evident than ever. They have the ability to inspire, provoke, and ignite change. Such is the essence of the book **Chemistry Physics Forces Study Guide**, a literary masterpiece that delves deep into the significance of words and their affect our lives. Published by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we will explore the book is key themes, examine its writing style, and analyze its overall effect on readers.

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