

**NEB CLASS 11**  
**MODEL QUESTION**  
**(EFFECTIVE FROM 2077)**

**Grade XI**

**Subject Code : Che. 301**

**Time :3 hours**

**Attempt all questions**

**Subject: Chemistry (Theory)**

**Full marks: 75**

**Group 'A'**

**Circle the best alternative to the following questions.**

**(11×1 = 11)**

1. How many atoms are there in two molecules of water?
  - a. 3
  - b. 4
  - c. 5
  - d. 6
2. What is the number of moles of ammonia gas formed when 0.5 mole of nitrogen gas is reacted with excess of hydrogen gas?
  - a. 0.5
  - b. 1
  - c. 2
  - d. 3
3. Which of the following bonding is responsible for the solubility of ammonia gas in water?
  - a. Hydrogen bonding
  - b. Ionic bonding
  - c. Covalent bonding
  - d. Van der Waals' force
4. What happens when Sulphur dioxide ( $\text{SO}_2$ ) gas is passed through an acidified solution of hydrogen sulfide ( $\text{H}_2\text{S}$ ) gas?
  - a.  $\text{SO}_2$  is oxidized to Sulphur
  - b.  $\text{H}_2\text{S}$  is reduced to Sulphur
  - c.  $\text{SO}_2$  is oxidized to  $\text{H}_2\text{SO}_4$
  - d.  $\text{SO}_2$  is reduced to Sulphur

# Chemistry Paper June Exam Grade11

**Daniela Niemeyer**



**Chemistry Paper June Exam Grade11:**

## Reviewing **Chemistry Paper June Exam Grade11**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is actually astonishing. Within the pages of "**Chemistry Paper June Exam Grade11**," an enthralling opus penned by a highly acclaimed wordsmith, readers set about an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve into the book's central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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### **Table of Contents Chemistry Paper June Exam Grade11**

1. Understanding the eBook Chemistry Paper June Exam Grade11
  - The Rise of Digital Reading Chemistry Paper June Exam Grade11
  - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Paper June Exam Grade11
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an eBook Chemistry Paper June Exam Grade11
  - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry Paper June Exam Grade11
  - Personalized Recommendations
  - Chemistry Paper June Exam Grade11 User Reviews and Ratings

- Chemistry Paper June Exam Grade11 and Bestseller Lists
- 5. Accessing Chemistry Paper June Exam Grade11 Free and Paid eBooks
  - Chemistry Paper June Exam Grade11 Public Domain eBooks
  - Chemistry Paper June Exam Grade11 eBook Subscription Services
  - Chemistry Paper June Exam Grade11 Budget-Friendly Options
- 6. Navigating Chemistry Paper June Exam Grade11 eBook Formats
  - ePub, PDF, MOBI, and More
  - Chemistry Paper June Exam Grade11 Compatibility with Devices
  - Chemistry Paper June Exam Grade11 Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Chemistry Paper June Exam Grade11
  - Highlighting and Note-Taking Chemistry Paper June Exam Grade11
  - Interactive Elements Chemistry Paper June Exam Grade11
- 8. Staying Engaged with Chemistry Paper June Exam Grade11
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Chemistry Paper June Exam Grade11
- 9. Balancing eBooks and Physical Books Chemistry Paper June Exam Grade11
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Chemistry Paper June Exam Grade11
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Chemistry Paper June Exam Grade11
  - Setting Reading Goals Chemistry Paper June Exam Grade11
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chemistry Paper June Exam Grade11
  - Fact-Checking eBook Content of Chemistry Paper June Exam Grade11
  - Distinguishing Credible Sources

13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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