SYLLABUS

Time Allowed: 3 Hours

Max. Marks: 30

Evaluation Scheme for Examination	Marks
Volumetric Analysis	08
Salt Analysis	08
Content Based Experiment	06
Project work	04
Class Record and Viva	04
Total	30

PRACTICAL SYLLABUS

Total Periods: 60

Micro-chemical methods are available for several of the practical experiments. Wherever possible, such techniques should be used.

A. Surface Chemistry

- a) Preparation of one lyophilic and one lyophobic sol.
 - Lyophilic sol-starch, egg albumin and gum
- Lyophobic sol-aluminum hydroxide, ferric hydroxide and arsenous sulphide
- b) Dialysis of sol-prepared in (α) above.
- Study of the role of emulsifying agents in stabilizing the emulsion of different oils.

B. Chemical Kineties

- a) Effect of concentration and temperature on the rate of reaction between Sodium Thiosulphate and Hydrochloric acid.
- b) Study of reaction rates of any one of the following:
- Reaction of Iodide ion with Hydrogen Peroxide at room temperature using different concentration of Iodide ions.
- (ii) Reaction between Potassium Iodate (KIO₃), and Sodium Sulphite (Na₂SO₃) using starch solution as indicator (clock reaction).

C. Thermochemistry

Any one of the following experiments:

- (i) Enthalpy of dissolution of Copper Sulphate or Potassium Nitrate.
- (ii) Enthalpy of neutralization of a strong acid (HCl) and a strong base (NaOH).
- (iii) Determination of enthalpy change during interaction (Hydrogen bond formation) between Acetone and Chloroform.

D. Electrochemistry

 a) Variation of cell potential in Zn/Zn²⁺ || Cu²⁺/Cu with change in concentration of electrolytes (CuSO₄ or ZnSO₄) at room temperature.

E. Chromatography

- Separation of pigments from extracts of leaves and flowers by paper chromatography and determination of R, value
- Separation of constituents present in an inorganic mixture containing two cations only (constituents having large difference in R_f values to be provided).

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Annelies Wilder-Smith

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