

Revision worksheet , Grade 11, Chemistry.

1. What is isomerism? Explain functional isomerism, metamerism and tautomerism with the help of suitable examples.
2. What do you mean by resonance. Explain with the help of an example.
3. What do you mean by Electron Displacement Effect? Explain inductive effect with the help of a proper example.
4. Write a note on the following reactions:-
 - a. Sabatier and Senderens reaction
 - b. Friedel Craft's reaction
 - c. Isomerisation
 - d. Dehydrohalogenation
 - e. Markonikov's rule
 - f. Ozonolysis
 - g. Decarboxylation
 - h. Wurtz reaction
 - i. Kolbe's electrolysis
 - j. Reforming or aromatization
 - k. Dehalogenation
 - l. Peroxide effect
 - m. Reaction of an alkene with KMnO_4
 - n. Dehydration
5. What is Grignard's reagent? How is it prepared? It was observed that Grignard's reagent should be stored in moisture proof containers. Why?
6. Identify the oxidizing agent, reducing agent, oxidant and reductant from the following using oxidation number method
$$\text{Zn} + 2\text{HCl} \longrightarrow \text{ZnCl}_2 + \text{H}_2$$
7. What is disproportionation reaction? Give an example.
8. What do you mean by an open system, closed system and an isolated system? Give an example.
9. What do you mean by saying that heat absorbed by a system is +ve and heat liberated by a system is -ve. Explain.
10. Explain the first law of Thermodynamics. Write its mathematical representation.
11. What do you mean by spontaneous and non spontaneous process. Give one example each.
12. State and explain Zeroth Law of Thermodynamics.
13. Explain Molarity, molality, mass percent and mole fraction.
14. Explain Aufbau principle, Hund's rule and Pauli's exclusion principle.
15. Draw the shapes of s, p and d orbitals.
16. What is ionization enthalpy? The ionization enthalpy of noble gases are very high. Why
17. What is electron gain enthalpy?

Chemistry Grade11 Revision 2013 Memo

Albert Holderness



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Food Processing Stephanie Clark, Stephanie Jung, Buddhi Lamsal, 2014-06-03 FOOD PROCESSING Food Processing Principles and Applications Second Edition is the fully revised new edition of this best selling food technology title Advances in food processing continue to take place as food scientists and food engineers adapt to the challenges imposed by emerging pathogens environmental concerns shelf life quality and safety as well as the dietary needs and demands of humans In addition to covering food processing principles that have long been essential to food quality and safety this edition of Food Processing Principles and Applications unlike the former edition covers microbial enzyme inactivation kinetics alternative food processing technologies as well as environmental and sustainability issues currently facing the food processing industry The book is divided into two sections the first focusing on principles of food processing and handling and the second on processing technologies and applications As a hands on guide to the essential processing principles and their applications covering the theoretical and applied aspects of food processing in one accessible volume this book is a valuable tool for food industry professionals across all manufacturing sectors and serves as a relevant primary or supplemental text for students of food science Revision Notes for Advanced Level Physical Chemistry David Gwyn Davies, Terence Victor George Kelly, 1976

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



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