

THE FOLLOWING SERVES AS ONLY A GUIDE

## CHEMICAL FORMULAS

1. What is the overall charge on a chemical formula? \_\_\_\_ 0 \_\_\_\_

2. ionic bonds form between: \_\_\_\_ metals and non-metals \_\_\_\_  
covalent bonds form between: \_\_\_\_ two or more non-metals \_\_\_\_

3-5 Write the formula for each of the following:

- sulfur trioxide \_\_\_\_  $\text{SO}_3$  \_\_\_\_
- nitrogen dioxide \_\_\_\_  $\text{N}_2\text{O}_4$  \_\_\_\_
- hydrofluoric acid: \_\_\_\_  $\text{HF}$  \_\_\_\_
- phosphoric acid \_\_\_\_  $\text{H}_3\text{PO}_4$  \_\_\_\_
- nitrous acid \_\_\_\_  $\text{HNO}_2$  \_\_\_\_
- nickel (II) carbonate \_\_\_\_  $\text{NiCO}_3$  \_\_\_\_
- silver hydroxide \_\_\_\_  $\text{AgOH}$  \_\_\_\_
- aluminum carbonate \_\_\_\_  $\text{Al}_2(\text{CO}_3)_3$  \_\_\_\_
- iron (II) oxide \_\_\_\_  $\text{Fe}_2\text{O}_3$  \_\_\_\_
- barium phosphate \_\_\_\_  $\text{Ba}_3(\text{PO}_4)_2$  \_\_\_\_

Write the name for each of the following compounds:

- $\text{N}_2\text{O}_5$  \_\_\_\_\_ Dinitrogen pentoxide
- $\text{S}_2\text{F}_6$  \_\_\_\_\_ Disulfur hexafluoride
- $\text{H}_2\text{SO}_4$  \_\_\_\_\_ Sulfuric acid
- $\text{HF}$  \_\_\_\_\_ Hydrofluoric acid
- $\text{HClO}_3$  \_\_\_\_\_ Chlorous acid
- $\text{Mn}(\text{OH})_2$  \_\_\_\_\_ Manganese (II) hydroxide
- $\text{Na}_2\text{S}$  \_\_\_\_\_ Sodium sulfide
- $\text{CaO}$  \_\_\_\_\_ Calcium oxide
- $\text{Sr}(\text{NO}_3)_2$  \_\_\_\_\_ Strontium nitrate
- $\text{Cu}_2\text{N}$  \_\_\_\_\_ Copper(I) nitride

6. Write the ion for nitrate, nitrite, and nitride  
\_\_\_\_  $\text{NO}_3^-$  \_\_\_\_  $\text{NO}_2^-$  \_\_\_\_  $\text{N}^{3-}$  \_\_\_\_

7. When are parentheses used in formula writing? When you have a polyatomic ion in the formula and the oxidation numbers are not the same and when there are more than one polyatomic ion necessary for stability in the formula.

8. What is the purpose for using Roman numerals? In other words, why must they be used with certain transition metals? Certain transition metals form more than one ion and to separate these from each other Roman numerals are used to indicate the charge of each form of the metal. Ex. Iron(II) and Iron(III).

## CHEMICAL EQUATION

### 1. BALANCING A CHEMICAL EQUATION

+ What do these symbols mean in a reaction?

|      |                  |
|------|------------------|
| +    | yields           |
| (l)  | liquid           |
| (s)  | solid            |
| (g)  | gas              |
| (aq) | aqueous solution |
| Δ    | heat             |

### 2. COEFFICIENTS & SUBSCRIPTS

+ How are chemical equations balanced?  
Using coefficients

### 3. BALANCE CHEMICAL EQUATIONS & TYPES OF RXN

+ What are the 5 types of reactions? (Briefly describe each one.)

- Synthesis
- Decomposition
- Single replacement
- Double replacement
- Combustion (exothermic and endothermic)

+ Balance the following reactions and identify the type



Type: \_\_\_\_\_  
Redox



Type: \_\_\_\_\_  
Synthesis



Type: \_\_\_\_\_  
Redox



Type: \_\_\_\_\_  
Decomposition



Type: \_\_\_\_\_  
Complete combustion

### 4. PREDICTING PRODUCTS IN A CHEMICAL EQUATION

+ Predict the products of the following reactions, then balance.



d) A combustion rxn involves the rxn of a hydrocarbon and \_\_\_\_  $\text{O}_2$  \_\_\_\_

e) What are the two products of a complete combustion rxn?  
\_\_\_\_  $\text{CO}_2$  \_\_\_\_ and \_\_\_\_  $\text{H}_2\text{O}$  \_\_\_\_

f) What are the two products of an incomplete combustion rxn?  
\_\_\_\_  $\text{CO}$  \_\_\_\_ and \_\_\_\_  $\text{H}_2\text{O}$  \_\_\_\_

### 5. LAW OF CONSERVATION OF MASS

+ In a chemical rxn what is true about the mass of the reactants compared to the mass of the products? \_\_\_\_ They must be equal \_\_\_\_

## STOICHIOMETRY

+ What is the first thing you must do to an equation before you begin a stoichiometric calculation? Balance it.

### 6. FINDING MOLES & MASS

+ What is the molar mass of  $\text{Ba}_3(\text{PO}_4)_2$ ?  
\_\_\_\_ 601.05 g \_\_\_\_

+ What is the molar mass of  $\text{O}_2$ ? \_\_\_\_ 32.00 g \_\_\_\_

# Chemistry Final Exam 2nd Semester

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*United States Air Force Academy* United States Air Force Academy,1974      **Curriculum Handbook with General Information Concerning ... for the United States Air Force Academy** United States Air Force Academy,2004

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In the last decade the development of new technologies has made innovation a fundamental pillar of education Teaching innovation includes the evolution of both teaching and learning models to drive improvements in educational methodologies Teaching innovation is a pioneer in the understanding and comprehension of the different teaching methodologies and models developed in the academic area Teaching innovation is a process that seeks validation in the academic and teaching communities at universities in order to promote the improvement and its practices and uses in the future characterized by digital development and data based methods Teaching Innovation in University Education Case Studies and Main Practices features the major practices and case studies of teaching innovation developed in recent years at universities It is a source on study cases focused on teaching innovation methodologies as well as on the identification of new technologies that will help the development of initiatives and practices focused on teaching innovation at higher education institutions Covering topics such as didactic strategics service learning and technology based gamification this premier reference source is an indispensable resource for pre service teachers lecturers students faculty administrators libraries entrepreneurs researchers and academicians

Making Chemistry Relevant Sharmistha Basu-Dutt,2010-02-19 Unique new approaches for making chemistry accessible to diverse students Students interest and achievement in academics improve dramatically when they make connections between what they are learning and the potential uses of that knowledge i n the workplace and or in the world at large Making Chemistry Relevant presents a unique collection of strategies that have been used successfully in chemistry classrooms to create a learner sensitive environment that enhances academic achievement and social competence of students Rejecting rote memorization the book proposes a cognitive constructivist philosophy that casts the teacher as a facilitator helping students to construct solutions to problems Written by chemistry professors and research groups from a wide variety of colleges and universities the book offers a number of creative ways to make chemistry relevant to the student including Teaching science in the context of major life issues and STEM professions Relating chemistry to current events such as global warming pollution and terrorism Integrating science research into the undergraduate laboratory curriculum Enriching the learning experience for students with a variety of learning styles as well as accommodating the visually challenged students Using media hypermedia games and puzzles in the teaching of chemistry Both novice and experienced faculty alike will find valuable ideas ready to be applied and adapted to enhance the learning experience of all their students

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**Resources in Education** ,1991-10 *Essential Advanced General Mathematics Third Edition Enhanced TIN/CP Version* Michael Evans,Kay Lipson,Douglas Wallace,Sue Avery,2011-04 The Essential VCE Mathematics series has a reputation for mathematical excellence with an approach developed over many years by a highly regarded author team of practising teachers and mathematicians This approach encourages understanding through a wealth of examples and exercises with an emphasis on VCE examination style questions New in the enhanced versions TI Nspire OS3 and Casio ClassPad calculator explanations examples and problems are integrated into the text Page numbers in the printed text reflect the previous TI nspire and Casio ClassPad version allowing for continuity and compatibility Digital versions of the student text are available in Interactive HTML and PDF formats through Cambridge GO

*University of Nebraska-Lincoln, Catalog: ARTS & SCIENCES, COLLEGE OF.* University of Nebraska--Lincoln. College Of Arts & Sciences,1928

**Essentials of Physical Chemistry** Don Shillady,2011-07-27 At a time when U S high school students are producing low scores in mathematics and science on international examinations a thorough grounding in physical chemistry should not be considered optional for science undergraduates Based on the author s thirty years of teaching Essentials of Physical Chemistry merges coverage of calculus with chemistry and molecular physics in a friendly yet thorough manner Reflecting the latest ACS guidelines the book can be used as a one or two semester course and includes special topics suitable for senior projects The book begins with a math and physics review to ensure all students start on the same level and then discusses the basics of thermodynamics and kinetics with mathematics tuned to a level that stretches students abilities It then provides material for an optional second semester course that shows students how to apply their enhanced mathematical skills in a brief historical development of the quantum mechanics of molecules Emphasizing spectroscopy the text is built on a foundation of quantum chemistry and more mathematical detail and examples It contains sample classroom tested exams to gauge how well students know how to use relevant formulas and to display successful understanding of key concepts Coupling the development of mathematical skills with chemistry concepts encourages students to learn mathematical derivations Mini biographies of famous scientists make the presentation more interesting from a people point of view Stating the basic concepts of quantum chemistry in terms of analogies provides a pedagogically useful technique Covering key topics such as the critical point of a van der Waals gas the Michaelis Menten equation and the entropy of mixing this classroom tested text highlights applications across the range of chemistry forensic science pre medical science and chemical engineering In a presentation of fundamental topics held together by clearly established mathematical models the book supplies a quantitative discussion of the merged science of physical chemistry

**Those Red Tag Bastards** Editor Don Shepperd,2012-08-08 A

collection of stories written by the members of the Class of 1962 the fourth class to graduate from the United States Air Force Academy the original Red Tag Bastards on the occasion of their 50th class reunion *Teach Students How to Learn* Sandra Yancy McGuire, 2023-07-03 Co published with NISOD Miriam a freshman Calculus student at Louisiana State University made 37.5% on her first exam but 83% and 93% on the next two Matt a first year General Chemistry student at the University of Utah scored 65% and 55% on his first two exams and 95% on his third These are representative of thousands of students who decisively improved their grades by acting on the advice described in this book What is preventing your students from performing according to expectations Sandra McGuire offers a simple but profound answer If you teach students how to learn and give them simple straightforward strategies to use they can significantly increase their learning and performance For over a decade Sandra McGuire has been acclaimed for her presentations and workshops on metacognition and student learning because the tools and strategies she shares have enabled faculty to facilitate dramatic improvements in student learning and success This book encapsulates the model and ideas she has developed in the past fifteen years ideas that are being adopted by an increasing number of faculty with considerable effect The methods she proposes do not require restructuring courses or an inordinate amount of time to teach They can often be accomplished in a single session transforming students from memorizers and regurgitators to students who begin to think critically and take responsibility for their own learning Sandra McGuire takes the reader sequentially through the ideas and strategies that students need to understand and implement First she demonstrates how introducing students to metacognition and Bloom's Taxonomy reveals to them the importance of understanding how they learn and provides the lens through which they can view learning activities and measure their intellectual growth Next she presents a specific study system that can quickly empower students to maximize their learning Then she addresses the importance of dealing with emotion attitudes and motivation by suggesting ways to change students' mindsets about ability and by providing a range of strategies to boost motivation and learning finally she offers guidance to faculty on partnering with campus learning centers She pays particular attention to academically unprepared students noting that the strategies she offers for this particular population are equally beneficial for all students While stressing that there are many ways to teach effectively and that readers can be flexible in picking and choosing among the strategies she presents Sandra McGuire offers the reader a step by step process for delivering the key messages of the book to students in as little as 50 minutes Free online supplements provide three slide sets and a sample video lecture This book is written primarily for faculty but will be equally useful for TAs tutors and learning center professionals For readers with no background in education or cognitive psychology the book avoids jargon and esoteric theory *Organic Chemistry Education Research into Practice* Jay Wackerly, Sarah Zingales, Michael Wentzel, Gautam Bhattacharyya, Brett McCollum, 2025-03-25 This Research Topic has three main goals 1 provide a platform for instructors of organic chemistry to showcase evidence based methods and educational theories they have utilized in their

classrooms 2 build new and strengthen existing connections between educational researchers and practitioners and 3 highlight how people have used chemical education based research in their teaching practice There are places in the literature dedicated for chemical education research CER however there is not a clear avenue for those that have changed their teaching methods based on published CER and report their experiences Creating this article collection will foster collaboration between chemical education researchers and teachers of organic chemistry This opportunity allows these instructors to share evidence based practices experiences challenges and innovative approaches from CER literature and beyond This Research Topic bridges discipline based education research and the scholarship of teaching and learning which will help advance organic chemistry education and improve student outcomes

**Announcements for Register of Alumni**  
University of Michigan. College of Pharmacy,1901 **Advances in Medical Education** A.J.J.A. Scherpbier,Cees P.M. van der Vleuten,J.J. Rethans,A.F.W. van der Steeg,2012-12-06 About 550 registrants from 51 different countries attended the Seventh Ottawa Conference on Medical Education and Assessment in Maastricht We received 525 abstracts for the conference divided in thematic poster sessions and platform presentations Organising the conference was an honour and we tried to meet the high standards of a friendly and relaxed atmosphere which has characterized previous Ottawa conferences During and after the conference about 250 papers were submitted for publication in the conference proceedings leaving us little time for a post conference depression Despite the large number of papers the editors have attempted to review and edit the papers as care fully as possible Occasionally however correspondence exceeded reasonable deadlines preventing careful editing of a small number of the papers Although we felt that our editorial task was not quite finished we nevertheless decided to include these papers We thank the many authors for their enthusiastic and prompt response to occasionally tedious editorial suggestions and requests We are sure that this collective effort has resulted in a book that will make an important contribution to the field of medical education The editors want to thank Jocelyn Flippo Berger whose expertise with desk top publishing and perseverance was a great help

Whole-class Inquiry Dennis W. Smithenry,Joan Gallager-Bolos,2009 In response to requests from science education professionals this is the perfect vehicle for implementing and assessing this concept of whole class inquiry in your classroom This is a must have package for preservice and inservice middle and high school science teachers

## **Chemistry Final Exam 2nd Semester** Book Review: Unveiling the Power of Words

In a global driven by information and connectivity, the energy of words has be much more evident than ever. They have the capability to inspire, provoke, and ignite change. Such is the essence of the book **Chemistry Final Exam 2nd Semester**, a literary masterpiece that delves deep into the significance of words and their affect our lives. Written by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we will explore the book is key themes, examine its writing style, and analyze its overall affect readers.

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