

Equilibria Test SCH4U #11 Name:

____K ____T ____C ____A Assessment of Learning
(Product)
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Multiple Choice: Circle the most correct answer. (10K)

1. At equilibrium, _____.

- A) all chemical reactions have ceased
- B) the rates of the forward and reverse reactions are equal
- C) the rate constants of the forward and reverse reactions are equal
- D) the value of the equilibrium constant is 1
- E) the limiting reagent has been consumed

2. Methanol, CH_3OH , can be produced by the following:



The conditions that are necessary to maximize the equilibrium yield of CH_3OH are:

- A) low temperature and low pressure
- B) high temperature and low pressure
- C) low temperature and high pressure
- D) high temperature and high pressure

3. The value for K_{eq} for a reaction is found to be 9.2×10^6 . What does this imply??

- A) products are favoured
- B) reactants are favoured
- C) reactant concentration = product concentration
- D) the system was open

4. The expression for K_{p} for the reaction below is _____.



- A) $\frac{P_{\text{CO}_2} P_{\text{H}_2\text{O}}}{P_{\text{CH}_4}}$
- B) $\frac{P_{\text{CO}_2} P_{\text{H}_2\text{O}}}{P_{\text{CuO}} P_{\text{CH}_4}}$
- C) $\frac{P_{\text{CO}_2} P_{\text{H}_2\text{O}}}{P_{\text{CH}_4}}$
- D) $\frac{P_{\text{CO}_2} P_{\text{H}_2\text{O}}}{P_{\text{CuO}}}$
- E) $\frac{P_{\text{CH}_4}}{P_{\text{CO}_2} P_{\text{H}_2\text{O}}}$

5. Assume that the reaction below is at equilibrium. The volume is decreased.



- A) shifts to the right
- B) No change occurs
- C) shifts to the left

6. Assume that the reaction below is at equilibrium. H_2S is removed. How will the system respond to the stress?



- A) shifts to the right
- B) No change occurs
- C) shifts to the left

7. The reaction:



Chemistry Equilibrium Test Answers

SA Adler



Chemistry Equilibrium Test Answers:

CliffsNotes AP Chemistry Bobrow Test Preparation Services, 2009-02-09 The book itself contains chapter length subject reviews on every subject tested on the AP Chemistry exam as well as both sample multiple choice and free response questions at each chapter's end Two full length practice tests with detailed answer explanations are included in the book

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