

Chemistry EOC Review – Answer Key

- 1) a. 68,200,000 cg b. 5000 mm c. 0.0125 kL d. 0.548 dm
- 2) ~21.5 mL
- 3) production of heat & light, formation of a gas, formation of a precipitate, color change
- 4) $33.64 \text{ g} / 4.28 \text{ ml} = 7.86 \text{ g/ml} = \text{Fe}$
- 5) m.p. $-183^{\circ}\text{C} = \text{CH}_4$ (methane)
- 6) H_2 – splint test (burning splint ignites hydrogen)
 O_2 – splint test (oxygen reignites glowing splint)
 H_2O – cobalt chloride paper (paper changes from blue to pink in the presence of water vapor)
 CO_2 – splint test (carbon dioxide extinguishes burning splint);
 limewater test (limewater turns cloudy when carbon dioxide is bubbled through it,
 producing CaCO_3)
- 7) The protons are positively charged particles found inside the nucleus along with the neutrons, which have no charge. The total number of these nucleons in an atom is the mass # / atomic mass of the atom. The electrons are the negatively charged particles found orbiting in "clouds" around the nucleus. Electrons can be gained by the atom to form anions or lost to form cations.
- 8) The thing that makes any element on the periodic table different from all other elements is the number of protons in its nucleus. Isotopes are atoms of the same type of element which have different masses because they have different numbers of neutrons.
- 9) a. $^{39}_{19}\text{K}^+$
 b. $^{57}_{26}\text{Fe}$
 c. $^{33}_{16}\text{S}^{2-}$
- 10)

<u>chemical symbol</u>	<u>atomic number</u>	<u>mass number</u>	<u># of protons</u>	<u># of electrons</u>	<u># of neutrons</u>	<u>net charge</u>
K^+	19	42	19	18	23	+1
Sr	38	88	38	38	50	0
F^-	9	20	9	10	11	-1
P	15	31	15	15	16	0
Ca^{2+}	20	42	20	18	22	+2

- 11) uranium – 238 $238 - 92 = 146$ neutrons
- 12) (1) 199.59g
 (2) They contain the same # of protons (they are the same element), but different #s of neutrons (electrons don't matter, b/c they have virtually no mass), & thus different masses.

Chemistry Eoc Study Guide Answer

VM Jensen



Chemistry Eoc Study Guide Answer:

Study Guide and Solutions Manual for Seager/Slabaugh's Chemistry for Today Seager/Slabaugh, 2004 The fifth edition of the Study Guide and Student Solutions Manual has been updated to reflect all of the changes to the text This ancillary tests the student on the learning objectives in each chapter and provides answers to all of the even numbered end of chapter exercises New additional activities have been added to include a review of each section of the chapter and a section entitled Tying It All Together with a Laboratory Application

Roadmap to the Virginia SOL Princeton Review, 2005 Roadmap to the Virginia SOL EOC Chemistry includes strategies that are proven to enhance student performance The experts at The Princeton Review provide content review of the crucial material most likely to appear on the test detailed lessons complete with test taking techniques for improving test scores 2 complete practice Virginia SOL EOC Chemistry tests

South Dakota Chemistry Eoc Success Strategies Study Guide South Dakota Eoc Exam Secrets Test Prep, 2014-08-22 South Dakota Chemistry EOC Success Strategies helps you ace the South Dakota End of Course Exams without weeks and months of endless studying Our comprehensive South Dakota Chemistry EOC Success Strategies study guide is written by our exam experts who painstakingly researched every topic and concept that you need to know to ace your test Our original research reveals specific weaknesses that you can exploit to increase your exam score more than you've ever imagined South Dakota Chemistry EOC Success Strategies includes The 5 Secret Keys to South Dakota EOC Success Time is Your Greatest Enemy Guessing is Not Guesswork Practice Smarter Not Harder Prepare Don't Procrastinate Test Yourself A comprehensive General Strategy review including Make Predictions Answer the Question Benchmark Valid Information Avoid Fact Traps Milk the Question The Trap of Familiarity Eliminate Answers Tough Questions Brainstorm Read Carefully Face Value Prefixes Hedge Phrases Switchback Words New Information Time Management Contextual Clues Don't Panic Pace Yourself Answer Selection Check Your Work Beware of Directly Quoted Answers Slang Extreme Statements Answer Choice Families Along with a complete in depth study guide for your specific South Dakota EOC exam and much more

Organic Chemistry T. W. Graham Solomons, Craig B. Fryhle, Scott A. Snyder, 2016-01-19 The 12th edition of Organic Chemistry continues Solomons Fryhle Snyder's tradition of excellence in teaching and preparing students for success in the organic classroom and beyond A central theme of the authors approach to organic chemistry is to emphasize the relationship between structure and reactivity To accomplish this the content is organized in a way that combines the most useful features of a functional group approach with one largely based on reaction mechanisms The authors philosophy is to emphasize mechanisms and their common aspects as often as possible and at the same time use the unifying features of functional groups as the basis for most chapters The structural aspects of the authors approach show students what organic chemistry is Mechanistic aspects of their approach show students how it works And wherever an opportunity arises the authors show students what it does in living systems and the physical world around us

Cracking the Virginia SOL Paul Foglino, Princeton Review (Firm), 2000

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Introduction to Organic Chemistry William H. Brown, Thomas Poon, 2016-01-13 *Introduction to Organic Chemistry* 6th Edition provides an introduction to organic chemistry for students who require the fundamentals of organic chemistry as a requirement for their major. It is most suited for a one-semester organic chemistry course. In an attempt to highlight the relevance of the material to students, the authors place a strong emphasis on showing the interrelationship between organic chemistry and other areas of science, particularly the biological and health sciences. The text illustrates the use of organic chemistry as a tool in these sciences; it also stresses the organic compounds both natural and synthetic that surround us in everyday life: in pharmaceuticals, plastics, fibers, agrochemicals, surface coatings, toiletry preparations, and cosmetics, food additives, adhesives, and elastomers. This text is an unbound three-hole punched version. Access to WileyPLUS sold separately.

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Michael L. Bishop, 2020-06-11 *Clinical Chemistry Principles Techniques and Correlations Enhanced Eighth Edition* demonstrates the how what why and when of clinical testing and testing correlations to help you develop the interpretive and analytic skills you ll need in your future career *A Practice-based Model of STEM Teaching* Alpaslan Sahin, 2015-07-21 The STEM Students on the Stage SOS TM model was developed by Harmony Public Schools with the goal of teaching rigorous content in an engaging fun and effective way In this book you will learn that the STEM SOS model is not only helping students learn STEM content and develop 21st century skills but also helping teachers improve their classroom climate through increased student teacher communication and a reduction in classroom management issues There are at least two ways in which this book is innovative First you will find student videos and websites associated with QR codes readers can use their QR readers to watch student videos related to the content in the chapter and see student e portfolio samples at their Google sites This provides the opportunity to see that what is discussed in the book actually happened Second the book is not about a theory it is an actual implemented model that has evolved through the years and has been used in more than 25 schools since 2012 Every year the model continues to be improved to increase its rigor and ease of implementation for both teachers and students In addition to using the book as a classroom teacher resource and guide it can also be used as a textbook in advanced graduate level curriculum and instruction educational leadership and STEM education programs Therefore STEM educators leaders pre service and in service teachers and graduate students will all benefit from reading this book Appendices will be one of the favorite aspects of this book for teachers who are constantly looking for ready to use student and teacher handouts and activities Full handouts including formative and summative assessments materials and grading rubrics will provide an opportunity for teachers and curriculum directors to understand the ideas and secrets behind the STEM SOS model Lastly STEM directors will find this to be one of the best STEM teaching model examples on the market because the model has fully accessible student and teacher handouts assessment materials rubrics and hundreds of student products e portfolios including video presentations and project brochures online **Chemistry and Metallurgy Research Building Replacement Project at Los Alamos National Laboratory**, 2003 **Exemplary Science in Grades 9-12**

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