#### Ouestions 17-19

- (A) NaC<sub>2</sub>H<sub>3</sub>O<sub>2</sub>
- (B) HC<sub>2</sub>H<sub>2</sub>O<sub>2</sub>
- (C) K(C)
- (D) NH<sub>0</sub>
- (E) HCI
- Is a salt that will undergo hydrolysis to form a basic solution.
- Will form a coordinate covalent bond with a hydronium ion.
- 19. Is a strong acid.

#### Questions 20-22

- (A)  $q = mc\Delta T$
- (B)  $q = H_{*}m$
- (C)  $P_1V_1 = P_2V_2$
- (D) D = m / V
- (E) K = C + 273

- Can be used to find the mass of an irregularly shaped solid.
- 21. Boyle's Law.
- Used to find energy gained or lost during a particular phase change.

#### Questions 23-25

- (A) Alpha particle
- (B) Beta particle
- (C) Neutron
- (D) Gamma ray
- (E) Positron
- 23. Has the greatest mass.
- 24. Has the greatest positive charge.
- 25. Has the same mass and charge as an electron.

# **Chemistry Diagnostic Test Topic 7 Answers**

Joseph A. Mascetta, Mark Kernion

#### **Chemistry Diagnostic Test Topic 7 Answers:**

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