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#### Chapter 12 – Stoichiometry Study Guide

The following example problems exhibit the types of calculations you will be expected to perform on tomorrow's test.

Describe a reaction in terms of puricles
Ex.) Describe the following reaction in terms of molecules

\* Note: sometimes you will need to include other descriptors such as atoms and formula same."

Describe a reaction in terms of moles
En.) Describe the following reaction in terms of moles

How do you determine the mass of the products for a balanced spection?
Ex.) What is the mass of the product formed in the following equation?

- How do you determine the volume of the seactants or products of a balanced seaction?
  - Ex.) What is the volume of the products and the resetants of the following equation?

 How do you show how a balanced reaction follows the Law of Conservation of Mass by showing that the mass on both sides of a reaction are equal?

Ex.) Show how the mans of the reactions equals the mans of the products for the following equation.

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John A. Olmsted, Gregory M. Williams

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