

# Marking scheme – 2017 (Compartment)

## CHEMISTRY (043)/ CLASS XII

Set 56/1/1

Q.No	Value Points	Marks
1	hcp	1
2	$AlCl_3 / Al^{3+}$	1
3	Orbital splitting energies are not sufficiently large for forcing pairing	1
4	2,3-dinitro phenol	1
5	Having no $\alpha$ -hydrogen	1
6	Vapour pressure of the solvent decreases in the presence of non – volatile solute (glucose) hence boiling point increases	2
7	(i) First order	1
	(ii) $s^{-1} / \text{time}^{-1}$	1
8	Hypophosphorous acid is a good reducing agent as it contains two P-H bonds. There is no P-H bond in orthophosphoric acid , so it is not a reducing agent Example : It reduces $AgNO_3$ to metallic silver/ chemical equation	1 1
	OR	
8	a) 4	1
	b) Due to lower bond dissociation enthalpy of $BiH_3$ as compared to $SbH_3$	1
9	i. Due to resonance the two O-O bond lengths are identical.	1
	ii. Due to strong bond formed by it with other elements.	1
10.	i) (b) is chiral	1
	ii) (a) will undergo $S_N2$ reaction faster	1
11	In bcc, $z=2$ ;  $d = (z \times M) / a^3 \times N_A$ (i) No. of atoms = $\frac{w}{M} \times N_A$ $2.5 \times 10^{24} = \frac{500g}{M} \times N_A$ $M = [500 \times N_A] / 2.5 \times 10^{24}$ (ii) Putting values of M in equation (i) $d = 2 \times 500 g \times N_A / [2.5 \times 10^{24} \text{ atoms} \times (400 \times 10^{-10} \text{ cm})^3 \times N_A]$ $d = 6.25 \text{ g/ cm}^3$ (or any other correct method )	  % 1  % 1

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