

Academic Chemistry 2nd Semester Final 2022

43. A solution in which the $[\text{OH}^-] = 1.0 \times 10^{-2} \text{ M}$ is:
a. acidic b. basic c. neutral d. none of the above
44. If the pH of a solution is 3.0, what is the $[\text{OH}^-]$?
a. 3 M b. 11 M c. $1 \times 10^{-3} \text{ M}$ d. $1 \times 10^{-11} \text{ M}$
45. If a solution has a pOH of 10.35 it is a(n):
a. acid b. base c. neutral solution
46. What is the pOH of a 0.55 M solution of HF?
a. 14 b. 3.70 c. 1.70 d. 3.41
47. What is the pH of a 0.19 M solution of HNO_3 ?
a. 0.72 b. 2.14 c. 4.88 d. 1.90

Chemistry 2nd Semester Finals

S Ben Porath



Chemistry 2nd Semester Finals:

Decoding **Chemistry 2nd Semester Finals**: Revealing the Captivating Potential of Verbal Expression

In an era characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its capability to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Chemistry 2nd Semester Finals**," a mesmerizing literary creation penned with a celebrated wordsmith, readers embark on an enlightening odyssey, unraveling the intricate significance of language and its enduring effect on our lives. In this appraisal, we shall explore the book's central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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