

# CHEMISTRY SPRING FINAL EXAM 2022-2023 WITH ACCURATE ANSWERS

atomic mass - Answer-the weighted average of the masses of the isotopes of an element

avogadro's number - Answer-the number of representative particles contained in one mole of a substance; equal to  $6.02 \times 10^{23}$

balanced equations - Answer-a chemical equation in which the mass is conserved; each side of the equation has the same number of atoms of each element

chemical equation - Answer-an expression representing a chemical reaction; the formulas of the reactants are connected by an arrow with the formulas for the products

double covalent bond - Answer-two atoms share two pairs of electrons

electron dot structure - Answer-notation that depicts valence electrons as dots around the atomic symbol of the element

electronegativity - Answer-ability of an atom to attract electrons when the atom is in a compound

empirical formula - Answer-a formula with the lowest whole number ratio of elements in a compound

hydrogen bond - Answer-attractive forces in which a hydrogen covalently bonded to an unshared electron pair of another electronegative atom

ionic bond - Answer-electrostatic attraction that binds oppositely charged ions together

ionization energy - Answer-the energy required to move an electron from an atom in its gaseous state

molar mass - Answer-a term used to refer to the mass of a mole of any substance

molar volume - Answer-volume occupied by one mole of gas at standard temperature and pressure

mole (mol) - Answer-amount of a substance that contains  $6.02 \times 10^{23}$  representative particles of that substance

# Chemistry Spring Final Exam Study Guide Answers

**Michal Rosen-Zvi**



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