UNIT 4: CHEMICAL REACTIONS STUDY GUIDE

a. double replacement

b. single replacement

OIAL T. OFICHIEOTE	KENETIONS STOOT	00100	
Multiple choice:			
C1. Chemical r	eactions		
a, simply chang	e the form or state of	of the reactants.	
b. create new	atoms.		
	ice new substances.		
d. create more	matter than what th	ey started with,	
B 2 Which of	the following is the co	orrect skeleton egua	tion for the
	zinc with liquid bromi		
a. Zn _{2(s)} + Br ₀		ne to produce some a	inc or online
b. Znoo + Brzo			
c. Zn(s) + Br2(s			
d. ZnBr₂(s) →			
A3. A substan		speed up a reaction l	but is not a product or
a. catalyst	b. coefficient	c. spectator ion	d. steroid
D 4 In a combi	ustion reaction, one of	f the reactants is al	ways
a a liquid	b water	c. a metal	d. oxygen
B5. Which of t simple binary compou a. The product b. The reactan c. The reactan	he following statemer	d. e. olecular substance.	
C6. The reaction?	ion 2P ₂ O ₅ → 4P + 50	D ₂ is an example of	which type of

c. decomposition

d. combustion

e. synthesis

Chemistry Section Review Study Guide Answer Key

Leo J. Malone, Theodore O. Dolter

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Groups II and VII Questions Chapter 14 Halogenoalkanes Questions Chapter 15 Hydrocarbons Questions Chapter 16 Introduction to Organic Chemistry Questions Chapter 17 Ionic Equilibria Questions Chapter 18 Lattice Energy Questions Chapter 19 Moles and Equations Questions Chapter 20 Nitrogen and Sulfur Questions Chapter 21 Organic and Nitrogen Compounds Questions Chapter 22 Periodicity Questions Chapter 23 Polymerization Questions Chapter 24 Rates of Reaction Questions Chapter 25 Reaction Kinetics Questions Chapter 26 Redox Reactions and Electrolysis Ouestions Chapter 27 States of Matter Ouestions Chapter 28 Transition Elements Ouestions The Alcohols and Esters Quiz Questions PDF e Book Chapter 1 interview questions and answers on Introduction to alcohols and alcohols reactions The Atomic Structure and Theory Quiz Questions PDF e Book Chapter 2 interview questions and answers on Atom facts elements and atoms number of nucleons protons electrons and neutrons The Benzene Chemical Compound 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electron pair repulsion theory enthalpy change of vaporization intermolecular forces ionic bonding ionic bonds and covalent bonds ionic bonds metallic bonding metallic bonding and delocalized electrons number of electrons sigma bonds and pi bonds sigma bonds pi bonds s orbital and p orbital Van der Walls forces and contact points The Chemistry of Life Quiz Questions PDF e Book Chapter 7 interview questions and answers on Introduction to chemistry enzyme specifity enzymes reintroducing amino acids and proteins The Electrode Potential Quiz Questions PDF e Book Chapter 8 interview questions and answers on Electrode potential cells and batteries E Plimsoll values electrolysis process measuring standard electrode potential quantitative electrolysis redox and oxidation The Electrons in Atoms Quiz Questions PDF e Book Chapter 9 interview guestions and answers on Electronic configurations electronic structure evidence ionization energy periodic table simple electronic structure sub shells and 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oxides electronegativity of period 3 oxides ionic bonds molecular structures of period 3 oxides oxidation number of oxides oxidation numbers oxides and hydroxides of period 3 elements oxides of period 3 elements period III chlorides periodic table electronegativity physical properties periodicity reaction of sodium and magnesium with water and relative melting point of period 3 oxides The Polymerization Quiz Questions PDF e Book Chapter 23 interview questions and answers on Types of polymerization polyamides polyesters and polymer deductions The Rates of Reaction Quiz Questions PDF e Book Chapter 24 interview questions and answers on Catalysis collision theory effect of concentration reaction

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