

Name _____ Date _____

Chapter 3 – Scientific Measurement Study Guide

- Qualitative vs. Quantitative measurements
- Scientific notation → decimal number
 - Ex.) 3.45×10^{-5}
 - Ex.) 6.02×10^8
- Decimal number → scientific notation
 - Ex.) 450,000,000
 - Ex.) .0000000045
- Multiply and divide numbers in scientific notation
 - What are the rules? What do you do with the coefficient? What do you do with the exponent?
 - Ex.) $(5 \times 10^{-2}) \times (11 \times 10^6)$
 - Ex.) $(8 \times 10^6) / (4 \times 10^5)$
- Add and subtract numbers in scientific notation
 - What are the rules? What do you do with the coefficient? What do you do with the exponent?
 - Ex.) $(4 \times 10^5) + (3 \times 10^5)$
 - Ex.) $(2 \times 10^2) - (4 \times 10^2)$
- % error (know how to use the formula)
 - Ex.) If the observed value for a density is 0.80 g/ml and the accepted value is 0.70 g/ml, what is the percent error?

Chemistry Scientific Measurement Study Guide Answers

J Ma



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Enjoying the Song of Term: An Emotional Symphony within **Chemistry Scientific Measurement Study Guide Answers**

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