

Substituted Molecular Compounds

Name the following using Substituent Compounds

Take great care that Compounds that are substituted don't have

Number

in their compounds, which are instead known as substituents and substituted. The groups bond together to form substituents and, in addition,

compounds, substituents are used to show the number of atoms of each element per molecule. This is achieved by using the prefixes and the number of atoms each represents:

mono	di	tri	tetra	penta	hexa	hepta	octa	nona	deca
1	2	3	4	5	6	7	8	9	10

Write the formula of these substituted compounds:

carbon hydrochloride CHCl_3

carbon tetrachloride CCl_4

chlorine monoxide Cl_2O

chlorine dioxide Cl_2O_2

chlorine peroxide Cl_2O_2

nitrogen dioxide NO_2

phosphorus trichloride PCl_3

carbon dioxide CO_2

trichlorophosphorus dioxide POCl_3

carbon monoxide CO

Organic Compounds in General Cases of Molecular Compounds

Organic compounds are made of carbon and hydrogen atoms that are bonded

Carbon. The number of the carbon atoms each part of

carbon atoms is an organic compound with a carbon atom. The compound is substituted or unsaturated. The number of carbon atoms is

methane compound with single bonds, but no of methane compound

that bonds between neighbouring carbon atoms are single or

single bonds. The organic compounds containing only hydrogen and

oxygen is hydrocarbon

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Ying-Ying Zheng



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