

Part A: The Density of Water

Experimental Data

	1 st Water Addition	2 nd Water Addition	3 rd Water Addition
Mass of Empty Cylinder	$W_1 = 110.909 \text{ gms}$	110.909 gms	110.909 gms
Mass of Cylinder + Water	132.467 g	154.831 gms	171.421 gms
Mass of Water only	21.558 g	43.922 gms	60.512 gms
Volume of Water	22.0 ml	44.8 ml	61.5 ml
Density of Water	21.558 g/ml	0.9813 g/ml	$\frac{60.512 \text{ gms}}{61.5 \text{ ml}} = 0.9839 \text{ g/ml}$
Average Density of Water	$= (0.9813 + 0.9859) \text{ g/ml} = 0.9819 \text{ g/ml}$		

Temperature of Water: 22°C

Data Analysis

- 1) Look up the true density of water at the temperature recorded: $22^\circ \text{C} = 0.9978$
 Obtain this value from http://physics.plymouth.edu/~johnson/courses/2012_Spring/Techniques/Exams/DensityOfWater-vs-Temp.pdf. Then use this to calculate the percent error in your average density of water. Show your work.

So percentage of Error =

$$\{(EV - TV) / TV\} \times 100$$

$$= \{(0.9819 - 0.9978) / 0.9978\} \times 100$$

Part B: The Density of Aluminum and the Thickness of Foil

Experimental Data

Table 1 - The Density of Aluminum

Mass of Empty Beaker	$(W_1) = 50.075 \text{ gms}$
Mass of Beaker and Al pellets	$(W_2) = 60.266 \text{ gms}$
Mass of Al pellets	$(W) = W_2 - W_1 = 60.266 - 50.075 = 10.191 \text{ gms}$
Initial volume of water in cylinder	$(V_1) = 32.9 \text{ ml}$
Final volume of water and Al pellets	$(V_2) = 36.2 \text{ ml}$
Volume of Al pellets	$(V) = 36.2 - 32.9 = 3.3 \text{ ml}$

Chemistry Lab Report Determining The Density Of Water Matriculation

**Jacqueline Barber, Laura Lowell, Marion
E. Buegler, Carolyn Willard**



Chemistry Lab Report Determining The Density Of Water Matriculation:

Drinking Water Chemistry Barbara Hauser, 2018-10-03 Whether you are a new employee or seasoned professional you need easy access to the latest test methods updated quality control procedures and calculations at your fingertips You need to perform analyses quickly and easily and troubleshoot problems as they arise You need a resource that is not only informative but also practical and easy to use *Drinking Water Chemistry A Laboratory Manual* fills this need The book gives you a thorough overview of the most basic and therefore important laboratory topics such as Laboratory Safety dos and don'ts based on real experience Sampling preservation techniques online sampling and record keeping Laboratory Instruments practical use ranges principles of operation calibration conditioning useful life and replacement common quality control issues Chemical Use reagents standards indicators purpose and use chemical quality and properties avoidance of contamination molecular weight calculations Quality Control replicate analyses spiked split and reference samples percent recovery of standard standard deviation control charts and everyday quality control measures Weights and Concentrations care and analytical balances mathematical conversions among concentration units dilutions and concentration changes The remaining chapters cover test analysis including reason for the test type of sample taken treatment plant control significance expected range of results appropriate quality control procedures apparatus used reagents including function concentration and instructions for preparation procedural steps calculations and notes on possible problems and references This is a working manual meant to be kept by your side in the lab not on the shelf in an office or library You can bend it you can lay it flat you can take it anywhere you do your job Useful and practical *Drinking Water Chemistry A Laboratory Manual* provides the information you need to perform tests understand the results apply them to the determination of water quality before and after treatment and troubleshoot any problems Laboratory Manual for the Examination of Water, Waste Water, and Soil Hans Hermann Rump, 1988 *Water Chemistry Laboratory Manual* David Albert Jenkins, American Association of Professors in Sanitary Engineering, 1971 **Determination and Verification of Required Water Chemistry Limits: Summary report**, 1984 Water Chemistry Laboratory Manual American Association of Professors in Sanitary Engineering, David Jenkins, 1972 *Laboratory Notes* Ellen H. Richards, 2016-08-03 Excerpt from *Laboratory Notes Sanitary Chemistry and Water Analysis* The application of chemistry to problems of public health or general sanitation is of comparatively recent date and its importance can hardly be overrated from the present standpoint Of the relation between health and the condition of air water and food materials as regards the action of the lower forms of life Soil and food are the hotbeds in which these lower forms of life are propagated and from which they are carried by water and air as well as by actual contact Sanitary biology is concerned with the organisms themselves sanitary chemistry with the detection and interpretation of the changes which they cause in the materials in which they grow and with the production of certain substances inimical to health In these notes are illustrated the more common changes which are now known to be caused by these organisms In

order to interpret correctly the results obtained it is necessary to know the normal composition of the materials in question. To determine this often involves analytical processes not strictly included under the head of sanitary tests namely the determination of total solids ash etc and certain changes of sanitary significance not due to the lower forms of life such as the vitiation of air by the burning of lights or the breathing of human beings. Sanitary chemistry may also properly include the examination of articles of food for the presence of adulterants or poisons. It is evident that any classification of topics is only temporary and tentative in the present state of knowledge. Water analysis and air analysis are each treated in a separate section. In the former the products of bacterial action are to be most carefully studied and accurately determined in order that the deductions from the results may be of value. Soil analysis has been fully described in works on agricultural chemistry see page and is therefore omitted. Food analysis is treated on the broad lines above indicated. A few typical substances only will be considered and the methods of examination given will be such as to elucidate the scientific aspect rather than technical detail of the subject. Both on account of its importance as a food stuff and on account of its availability for the tests milk has been chosen as a type of animal food. This class includes meats and meat products meat tablets peptones and other prepared animal foods. The same methods of examination may be applied to the analysis of fertilizers. About the Publisher Forgotten Books publishes hundreds of thousands of rare and classic books. Find more at www.forgottenbooks.com. This book is a reproduction of an important historical work. Forgotten Books uses state of the art technology to digitally reconstruct the work preserving the original format whilst repairing imperfections present in the aged copy. In rare cases an imperfection in the original such as a blemish or missing page may be replicated in our edition. We do however repair the vast majority of imperfections successfully. Any imperfections that remain are intentionally left to preserve the state of such historical works.

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