CHEMISTRY MID-TERM STUDY GUIDE -- CONSWERS

For questions 1-6, determine the number of significant figures in each measurement.

__4__1) 4658 g ___2_3) 0.0020 km ___1_5) 200 m

__4__2) 360.0 cm __3__4) 7.02×10⁵ mm ___4_6) 8050. S

For questions 7-10, determine the number of significant figures in the calculated answers without solving the problems.

__3___7) 4.033 X 8.99 = __3___9) 34.6 + 22.22 =

__d___11) In the measurement 305.67q, which digit is estimated?

a. The 3 b. The 0 c. The 6 d. The 7

__d___12) A piece of paper measures 220.5 cm by 76.3 cm. What is the area of this piece of paper expressed to the correct number of sig. figs.?

a. 16824.15 cm² b. 16824 cm² c. 16820 cm² d. 16800 cm²

______13) A sample of metal is found to have a density of 8.44 g/mL and a mass of 10.00g. What is the volume of this metal?

a. 1.18 mL b. 1.180 mL c. 0.844 mL d. 0.84 mL

___b__14) Calculate the density of a gas that has a volume of 1.033×10-2 L and a mass of 4.22×10-4 g.

a. 4.09x10²g/L
b. 4.09x10⁻²g/L
c. 2.45x10¹ L/g

For questions 15 - 18 choose "A" if the statement refers to an ionic compound and "B" if the statement refers to a covalent compound.

a. 2 b. 4 c. 5 d. 7

Chemistry Honors Midterm Study Guide Answers

James E. Brady

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