

ACID & BASE CHEMISTRY

STRONG ACIDS

HClO_4	Perchloric Acid
HClO_3	Chloric Acid
H_2SO_4	Sulfuric Acid
HNO_3	Nitric Acid
H_3O^+	Hydrodium ion
HI	Hydroiodonic Acid
HBr	Hydrobromic Acid

WEAK ACIDS

H_3PO_4	Phosphoric Acid
HNO_2	Nitrous Acid
HF	Hydrofluoric Acid
CH_3COOH	Ethanoic Acid
H_2CO_3	Carbonic Acid
H_2S	Hydrogen Sulfide
HCOOH	Methanoic Acid
H_2SO_3	Sulfurous Acid

STRONG BASES

NaOH	Sodium Hydroxide
LiOH	Lithium Hydroxide
KOH	Potassium Hydroxide
$\text{Ca}(\text{OH})_2$	Calcium Hydroxide
$\text{Ba}(\text{OH})_2$	Barium Hydroxide
Na_2CO_3	Sodium Carbonate

WEAK BASES

NH_3	Ammonia
NaHCO_3	Sodium hydrogen carbonate
CH_3NH_2	Methylamine

TYPES OF NEUTRALISATION REACTIONS

Hydroxide Base + Acid \rightarrow Salt + Water

Carbonate Base + Acid \rightarrow Carbon dioxide + Salt + Water

Ammonia + Acid \rightarrow Ammonium ion + Anion (from Acid)



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Laboratories Food and Agriculture Organization of the United Nations, 1998-01-01 **Pp/Chemistry** BarCharts, Inc., 2008-06-18 MSCEIS 2019 Lala Septem Riza, Eka Cahya Prima, Toni Hadibarata, Peter John Aubusson, 2020-07-30 The 7th Mathematics Science and Computer Science Education International Seminar MSCEIS was held by the Faculty of Mathematics and Natural Science Education Universitas Pendidikan Indonesia UPI and the collaboration with 12 University associated in Asosiasi MIPA LPTK Indonesia AMLI consisting of Universitas Negeri Semarang UNNES Universitas

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Table of Contents Chemistry Guide Acids And Bases Answe

1. Understanding the eBook Chemistry Guide Acids And Bases Answe
 - The Rise of Digital Reading Chemistry Guide Acids And Bases Answe
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Guide Acids And Bases Answe
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemistry Guide Acids And Bases Answe
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry Guide Acids And Bases Answe

- Personalized Recommendations
 - Chemistry Guide Acids And Bases Answer User Reviews and Ratings
 - Chemistry Guide Acids And Bases Answer and Bestseller Lists
5. Accessing Chemistry Guide Acids And Bases Answer Free and Paid eBooks
 - Chemistry Guide Acids And Bases Answer Public Domain eBooks
 - Chemistry Guide Acids And Bases Answer eBook Subscription Services
 - Chemistry Guide Acids And Bases Answer Budget-Friendly Options
 6. Navigating Chemistry Guide Acids And Bases Answer eBook Formats
 - ePub, PDF, MOBI, and More
 - Chemistry Guide Acids And Bases Answer Compatibility with Devices
 - Chemistry Guide Acids And Bases Answer Enhanced eBook Features
 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chemistry Guide Acids And Bases Answer
 - Highlighting and Note-Taking Chemistry Guide Acids And Bases Answer
 - Interactive Elements Chemistry Guide Acids And Bases Answer
 8. Staying Engaged with Chemistry Guide Acids And Bases Answer
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemistry Guide Acids And Bases Answer
 9. Balancing eBooks and Physical Books Chemistry Guide Acids And Bases Answer
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemistry Guide Acids And Bases Answer
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Chemistry Guide Acids And Bases Answer
 - Setting Reading Goals Chemistry Guide Acids And Bases Answer
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Chemistry Guide Acids And Bases Answer

- Fact-Checking eBook Content of Chemistry Guide Acids And Bases Answe
- Distinguishing Credible Sources

13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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