

CP Chemistry
Final Exam Review Sheet Fall 2010 ANSWER KEY

The multiple choice final exam is 15% of your overall final grade. To study for the multiple choice portion, it is suggested you use this review sheet, your notes, AKS log book, handouts, textbook, and labs. On the day of the exam, you will need to bring a #2 pencil and a calculator. A periodic table will be provided for you.

Lab Equipment/Safety

1. a) What is a hypothesis? testable, refutable explanation for an observation or a problem
b) What is a theory? A hypothesis that has been supported by repeated experiments
c) What is a law? it describes relationships in nature without exception
2. What is the difference between accuracy and precision? Accuracy means that the measurements made are correctly (they are very close to the accepted measurement). Precision means that a series of measurements are close to each other (you can make the same measurement more than once and get about the same measurement each time). Note: measurements can be precise and not accurate.



Measurement and Matter

3. Use the data table below to discern whether the students' measurements are accurate only, precise only, both accurate and precise, or neither accurate nor precise.

| | Measurement 1 | Measurement 2 | Measurement 3 | Measurement 4 |
|--------|---------------|---------------|---------------|---------------|
| Jameel | 5.0 g | 4.9 g | 5.0 g | 5.1 g |
| Cody | 8.0 g | 8.2 g | 8.0 g | 7.9 g |
| Rachel | 2.0 g | 12.0 g | 8.0 g | 20.0 g |
| Juan | 10.0 g | 9.9 g | 10.1 g | 9.9 g |

****The accepted value of the rock's mass is 5.0 g

- a) Describe Jameel's measurements. accurate & precise
 - b) Describe Cody's measurements. precise
 - c) Describe Rachel's measurements. neither
 - d) Describe Juan's measurements. precise
4. a) What piece of lab equipment measures mass? electronic balance
b) What piece of lab equipment most accurately measures volume of a liquid? graduated cylinder
c) What is the SI unit to measure length? meter Mass? gram
 5. a) What is the formula for density? $D = m/v$ (density = mass/volume)
 6. a) How is the volume of a regularly-shaped object, such as a box or book, found? $v = \text{length} \times \text{width} \times \text{height}$
b) What is the name for the method by which the volume of an irregularly-shaped object is found? water displacement method (subtract the final volume from the initial volume: $V = V_{\text{final}} - V_{\text{initial}}$)



Chemistry Final Exam Comprehensive Review Guide

Patrick Schnider



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