# CHAPTER 15 STUDY GUIDE

### **Energy and Chemical Change**

#### Section 15.1 Energy

In your textbook, read about the nature of energy.

In the space at the left, write true if the statement is true; if the statement is false, change the italicized word or phrase to make it true.

- Energy is the ability to do work or produce heat.

   The law of conservation of energy states that energy can be created and destroyed.
  - Chemical potential energy is energy stored in a substance because of its composition.
  - 4. Heat is a form of energy that flows from a warmer object to a cooler object.
  - A calorie is the amount of energy required to raise the temperature of one graw of pure water by one degree Celsius.
  - 6. A calorie is the SI unit of heat and energy.
  - The specific heat of a substance is the amount of heat required to raise the temperature of one gram of that substance by one degree Celsius.
  - 8. Kinetic energy is energy of motion.
  - Chemicals participating in a chemical reaction contain only potential energy.
  - 10. One nutritional Calorie is equal to 100 calories.
  - 11. One calorie equals 4.184 joules.
  - When a fuel is burned, some of its chemical potential energy is lost as heat.
  - To convert kilojoules to joules, divide the number of kilojoules by1000 joules/1 kilojoule.

#### Answer the following question. Show all your work.

 If the temperature of a 500.0-g sample of liquid water is raised 2.00°C, how much heat is absorbed by the water? The specific heat of liquid water is 4.184 J/(g-°C).

# **Chemistry Energy Study Guide Key**

David R. Klein

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Basic Concepts of Chemistry, 9e Study Guide and Solutions Manual Leo J. Malone, Theodore O. Dolter, 2012-01-03 The 9th edition of Malone's Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections assessment exercises at the end each section and relevant chapter problems at the end of each chapter A new Math Check allows quick access to the needed basic skill The first chapter now includes brief introductions to several fundamental chemical concepts and Chapter Synthesis Problems have been added to the end of each chapter to bring key concepts into one encompassing problem Every concept in the text is clearly illustrated with one or more step by step examples Making it Real essays have been updated to present timely and engaging real world applications emphasizing the relevance of the material they are learning This edition continues the end of chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter "O" Level Study Guide - 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