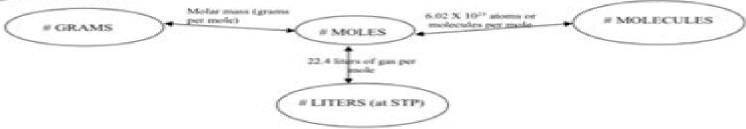
## CHEMISTRY B- MOLES PACKET NAME: PAGE 9 CHEMISTRY WORKSHEET # 6 MIXED MOLE PROBLEMS (GRAMS, MOLECULES, AND LITERS)

You now know three things a mole can be: a molar mass, 6.02 x 10<sup>23</sup> molecules and, for a gas, 22.4 liters at STP. We can use this information to convert grams to molecules or liters, molecules to grams or liters, or liters to grams or molecules.



EXAMPLE 1: What would be the volume in liters of 40.36 grams of neon at STP?

EXAMPLE 2: How many molecules would there be in 56 liters of carbon dioxide at STP?

#### SOLVE THE FOLLOWING PROBLEMS ON A SEPARATE SHEET OF PAPER.

- YOU MUST USE COMPLETE AND PROPER SET-UPS.
- SHOW THE MOLAR MASS CALCULATION WHENEVER THE PROBLEM REQUIRES YOU TO DO ONE. INCLUDE UNITS
- CIRCLE YOUR FINAL ANSWER.
- What would be the volume, in liters, of 85.5 grams of carbon monoxide at STP?
- How many molecules would there be in 0.500 grams of carbon disulfide?
- What would be the mass, in grams, of 45.0 liters of nitrogen at STP?
- 4. How many molecules of hydrogen are in a balloon full of hydrogen with a volume of 5.34 liters at STP?
- Your monning buys you a helium balloon at the circus. It has a volume of 4.00 liters at STP. What mass of helium, expressed in grams, does this balloon contain?
- How many molecules of ammonia would there be in 40.0 grams of ammonia?
- 7. What would be the mass, in grams, of 3.50 x 10<sup>25</sup> molecules of chlorine?
- What volume, expressed in liters, would 50.0 grams of fluorine occupy at STP?
- How many grams of oxygen would there be in 1.00 liter of oxygen at STP?

#### Extra Challenge:

10. How many molecules of water are there in 10 lbs of water?

# **Chemistry B Moles Packet Answers**

**Ralph Thomas Myers** 

#### **Chemistry B Moles Packet Answers:**

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Chemistry in the Community (ChemCom) American Chemical Society, 2011-06-17 Touted as the most successful NSF funded project published Chemistry in the Community ChemCom by the American Chemical Society ACS offers a meaningful and memorable chemistry program for all levels of high school students ChemCom covers traditional chemistry topics within the context of societal issues and real world scenarios Centered on decision making activities where students are responsible for generating data in an investigating analyzing that data and then applying their chemistry knowledge to solve the presented problem The text is intensively laboratory based with all 39 of the investigations integrated within the text not separate from the reading With the ChemCom program students learn more organic and biochemistry more environmental and industrial chemistry and more on the particulate nature of matter than other textbooks all within the relevance of solving problems that arise in everyday life Meticulously updated to meet the needs of today s teachers and students the new sixth edition of ChemCom adheres to the new science framework as well as the forthcoming next generation of science standards Incorporating advances in learning and cognitive sciences ChemCom's wide ranging coverage builds upon the concepts and principles found in the National Science Education Standards Correlations are available showing how closely aligned ChemCom is to these and other state standards Educart ICSE Class 10 One-shot Question Bank 2026 Chemistry (strictly for 2025-26 boards) Sir Tarun Rupani, 2025-07-12 Fast track your Chemistry revision with this exam ready resource This One shot Question Bank by Sir Tarun Rupani is designed to help ICSE Class 10 students revise the complete Chemistry syllabus quickly and thoroughly It simplifies theory boosts numerical accuracy and ensures strong exam practice all aligned with the

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