

# ORGO ACIDS & BASES

## CHEAT SHEET

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Arrhenius

Acid = Litmus Red  
 $H^+$  in  $H_2O$

Base Litmus Blue  
 $OH^-$  in  $H_2O$

Bronsted-Lowry

$H^+$  donor  
 ex.  $HCl$

Accepts  $H^+$   
 ex.  $NaOH$

Lewis (electrons)

Accepts  $e^-$  pair  
 ex.  $AlCl_3$

$e^-$  pair donor  
 ex.  $Cl^-$

Acid - Base Reaction



Equations to recognize

$$K_a = \frac{[H^+][A^-]}{[HA]} \quad pK_a = -\log(K_a) \rightarrow K_a \propto [H^+] \propto \frac{1}{pK_a}$$

Strong Acid

$\uparrow [H^+]$

$\uparrow K_a$

$\downarrow pK_a$

Weak Acid

$\downarrow [H^+]$

$\downarrow K_a$

$\uparrow pK_a$

$$pH = -\log [H^+] \\ pOH = -\log [OH^-]$$

$$pH + pOH = 14 \\ pK_a + pK_b = pK_w = 14$$

pH scale  
 $pH \rightarrow$   
 $\leftarrow pOH$

$pH = 1$   
 Acidic  
 $pOH = 14$

$pH = 7$   
 Neutral  
 $pOH = 7$

$pH = 14$   
 Basic  
 $pOH = 1$

STRONG ACIDS FORM STABLE CONJUGATE BASES

CARIO

C = charge of acid or conj. base

A = Atom holding charge

R = Resonance

I = Inductive effect

O = orbital / hybridization

See videos for full explanation + examples [LEAH4SCI.COM/acidbase](http://LEAH4SCI.COM/acidbase)

Charge:  $+ / \delta^+$  more acidic } When compare species  
 $- / \delta^-$  more basic }

Atom: In same period  $\uparrow$  eneg  $\uparrow$  acidity  
 In same group  $\uparrow$  size  $\uparrow$  acidity



Resonance:  $\uparrow$  Res  $\uparrow$  charge distribution  
 $\uparrow$  acidity (Aromaticity =  $\uparrow$  stable)

Inductive Effect:  $\uparrow$  e-neg nearby atom  
 $\uparrow$  acidity

Orbital / hybridization  $\uparrow$  % s  $\uparrow$  acidity  
 $sp = 50\% > sp^2 = 33\% > sp^3 = 25\%$

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# Chemistry Acid Bases And Salts Study Guide

**Rob Ritchie**



## Chemistry Acid Bases And Salts Study Guide:

Basic Concepts of Chemistry, 9e Study Guide and Solutions Manual Leo J. Malone, Theodore O. Dolter, 2012-01-03 The 9th edition of Malone's Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment. New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections, assessment exercises at the end of each section, and relevant chapter problems at the end of each chapter. A new Math Check allows quick access to the needed basic skill. The first chapter now includes brief introductions to several fundamental chemical concepts, and Chapter Synthesis Problems have been added to the end of each chapter to bring key concepts into one encompassing problem. Every concept in the text is clearly illustrated with one or more step-by-step examples. Making it Real essays have been updated to present timely and engaging real-world applications, emphasizing the relevance of the material they are learning. This edition continues the end-of-chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter.

*Barron's Science 360: A Complete Study Guide to Chemistry with Online Practice* Barron's Educational Series, Mark Kernion, Joseph A. Mascetta, 2021-09-07 Barron's Science 360 Chemistry is your complete go-to guide for everything chemistry. This comprehensive guide is an essential resource for High School and college courses, Homeschooling, Virtual Learning, Learning pods. Inside you'll find Comprehensive Content Review. Begin your study with the basic building block of chemistry and build as you go. Topics include atomic structure, chemical formulas, electrochemistry, the basics of organic chemistry, and much more. Effective Organization. Topic organization and simple lesson formats break down the subject matter into manageable learning modules that help guide a successful study plan customized to your needs. Clear Examples and Illustrations. Easy-to-follow explanations, hundreds of helpful illustrations, and numerous step-by-step examples make this book ideal for self-study and rapid learning. Practice Exercises. Each chapter ends with practice exercises designed to reinforce and extend key skills and concepts. These checkup exercises, along with the answers and solutions, will help you assess your understanding and monitor your progress. Access to Online Practice. Take your learning online for 50 practice questions designed to test your knowledge with automated scoring to show you how far you have come.

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*Robinson Chemistry Study Guide* Robinson,1992 *Study Guide for General Chemistry and College Chemistry, Eighth Editions by Holtzclaw and Robinson* Norman E. Griswold, Henry Fuller Holtzclaw,1988 "O"

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