Last Name:

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#### Chemistry 104 Assignment #9 Answer Key

- Describe the characteristic electron-domain geometry of each of the following numbers of electron domains about a central atom: a) 3, b) 4, c) 5, d) 6.
- a) trigonal planar b) tetrahedral c) trigonal bipyramidal
- d) oetahedral
- 2. Give the electron-domain and molecular geometries of a molecule that has the following electron domains on its central atom: a) four bonding domains and no nonbonding domains, b) three bonding domains and two nonbonding domains, c) five bonding domains and one nonbonding domain.

a) tetrahedral e.d.q. & m.q. b) trigonal bipyramidal e.d.g.; T-shaped mg. c) octahedral e.d.g.; square pyramidal mg.

Predict whether each of the following molecules is polar or non-polar: a) IF, b) CS2,

a) 
$$\vec{I} - \vec{F}$$
: polar b)  $\vec{S} = \vec{C} = \vec{S}$  non-polar c)

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